

### Boyle's Law

- a. If the \_\_\_\_\_ of a fixed \_\_\_\_\_ of gas varies \_\_\_\_\_ with the \_\_\_\_\_ of volume \_\_\_\_\_.
- b. \_\_\_\_\_
- c. \_\_\_\_\_

### Ideal Gas Theory and Boyle's Law

- a. \_\_\_\_\_ of a gas is caused by the \_\_\_\_\_ of the gas \_\_\_\_\_ themselves & the \_\_\_\_\_.
- b. If the gas is \_\_\_\_\_, the \_\_\_\_\_ of the volume of the \_\_\_\_\_, assuming \_\_\_\_\_, the pressure \_\_\_\_\_.
- c. \_\_\_\_\_, assuming \_\_\_\_\_, the \_\_\_\_\_ of the walls of the \_\_\_\_\_.
- d. \_\_\_\_\_ of the gas will \_\_\_\_\_.

Ex. 1. A certain quantity of gas has a volume of 0.01 m<sup>3</sup> at standard atmospheric pressure. After the volume is reduced, it exerts an absolute of 0.25 bar where the pressure is only 0.1013 bar. What is the volume of the gas at this absolute?

Ex. 2. A certain quantity of \_\_\_\_\_, see Fig. 1, exerts a gas that is volume of \_\_\_\_\_ m<sup>3</sup>. If the pressure is increased until the volume is \_\_\_\_\_ m<sup>3</sup>, what is the new pressure, assuming the temperature remains constant?

# Note Taking Guide Episode 902 Answers

**Mark Steyvers**



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