Now, balance the first reaction the REDON way! <u>Underline</u> the reducing agent and <u>Circle</u> the exidizing agent.

MINITED TO THE TOTAL CONTROL OF THE STATE OF

Learning Check

Why can't a REDOX reaction a be balanced the normal way?

- Maid Millian Millian Millian



In a REDOX reaction the number of atoms and the number of electrons transferred needs to be balanced. In a normal reaction, only the number of atoms needs to be balanced. There is no transfer of electrons. As a result, the normal way of balancing wen't work for a REDOX reaction. The electrons aren't being balanced.

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LM Reder

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