

Chapter 1: Limiting Reagents and Percent Yield (Pages 1-4)

Remember: Limiting reagent is the reactant that will completely run out first in a chemical reaction.
Calculate: The limiting reagent is the reactant that will completely run out first in a chemical reaction. The limiting reagent is the reactant that will completely run out first in a chemical reaction.

Remember: Limiting reagent is the reactant that will completely run out first in a chemical reaction.
 Calculate: The limiting reagent is the reactant that will completely run out first in a chemical reaction.



The reaction is balanced. The limiting reagent is the reactant that will completely run out first in a chemical reaction.

2H ₂	20.0 g/mol
O ₂	32.0 g/mol
2H ₂ O	36.0 g/mol
H ₂	2.0 g/mol

1. 20.0 g of H₂ and 32.0 g of O₂ are mixed. The limiting reagent is the reactant that will completely run out first in a chemical reaction.



2. 20.0 g of H₂ and 32.0 g of O₂ are mixed. The limiting reagent is the reactant that will completely run out first in a chemical reaction.

3. 20.0 g of H₂ and 32.0 g of O₂ are mixed. The limiting reagent is the reactant that will completely run out first in a chemical reaction.



4. 20.0 g of H₂ and 32.0 g of O₂ are mixed. The limiting reagent is the reactant that will completely run out first in a chemical reaction.



Percent Yield And Limiting Reactants Quiz

**Barron's Educational Series, Mark
Kernion, Joseph A. Mascetta**



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