

Unit 7: Balancing Chemical Reactions • Worksheet 2

Balance the following equations by inserting the proper coefficients.

1. $\text{_2_SO}_2 + \text{_1_O}_2 \rightarrow \text{_2_SO}_3$
 2. $\text{_2_CH}_4 + \text{_3_O}_2 \rightarrow \text{_2_CO} + \text{_4_H}_2\text{O}$
 3. $\text{_2_P} + \text{_3_Cl}_2 \rightarrow \text{_2_PCl}_3$
 4. $\text{_2_CO} + \text{_1_O}_2 \rightarrow \text{_2_CO}_2$
 5. $\text{_2_CH}_4 + \text{_1_O}_2 \rightarrow \text{_2_CH}_3\text{OH}$
 6. $\text{_2_Li} + \text{_1_Br}_2 \rightarrow \text{_2_LiBr}$
 7. $\text{_2_Al}_2\text{O}_3 \rightarrow \text{_4_Al} + \text{_3_O}_2$
 8. $\text{_2_Na} + \text{_2_H}_2\text{O} \rightarrow \text{_2_NaOH} + \text{_1_H}_2$
 9. $\text{_6_CO}_2 + \text{_6_H}_2\text{O} \rightarrow \text{_1_C}_6\text{H}_{12}\text{O}_6 + \text{_6_O}_2$
 10. $\text{_1_H}_2\text{SO}_4 + \text{_2_NaCl} \rightarrow \text{_2_HCl} + \text{_1_Na}_2\text{SO}_4$
- Two reactions used to get rid of sulfur dioxide, a pollutant from burning coal:
11. $\text{_3_H}_2 + \text{_1_SO}_2 \rightarrow \text{_1_H}_2\text{S} + \text{_2_H}_2\text{O}$
 12. $\text{_2_CaCO}_3 + \text{_2_SO}_2 + \text{_1_O}_2 \rightarrow \text{_2_CaSO}_4 + \text{_2_CO}_2$
-
13. $\text{_2_AgNO}_3 + \text{_1_CaCl}_2 \rightarrow \text{_2_AgCl} + \text{_1_Ca(NO}_3)_2$
 14. $\text{_2_HCl} + \text{_1_Ba(OH)}_2 \rightarrow \text{_1_BaCl}_2 + \text{_2_H}_2\text{O}$
 15. $\text{_1_H}_3\text{PO}_4 + \text{_3_NaOH} \rightarrow \text{_1_Na}_3\text{PO}_4 + \text{_3_H}_2\text{O}$
 16. $\text{_1_Pb(NO}_3)_2 + \text{_2_KI} \rightarrow \text{_1_PbI}_2 + \text{_2_KNO}_3$
 17. $\text{_3_CuO} + \text{_2_NH}_3 \rightarrow \text{_1_N}_2 + \text{_3_Cu} + \text{_3_H}_2\text{O}$
 18. $\text{_1_C}_2\text{H}_5\text{OH} + \text{_3_O}_2 \rightarrow \text{_2_CO}_2 + \text{_3_H}_2\text{O}$
 19. $\text{_2_C}_2\text{H}_6 + \text{_3_O}_2 \rightarrow \text{_2_CH}_3\text{COOH} + \text{_2_H}_2\text{O}$
 20. $\text{_3_NO}_2 + \text{_1_H}_2\text{O} \rightarrow \text{_2_HNO}_3 + \text{_1_NO}$

Rearranging Atoms Worksheet

Clifford Lane

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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