

10.4 Electrophilic Reactions

General Introduction An electrophile is an electron-seeking species, reacting with a lone pair or a double bond.

Electrophilic Reactions

Electrophilic reactions involving alkenes are addition reactions where both ends of the

- alkene are attacked by an electrophile (E⁺) to form a carbocation intermediate (an alkyl cation).
- The carbocation is then attacked by a nucleophile (Nu⁻) to form the final product, an alkylated alkane.

Electrophilic reactions involving alkenes are addition reactions where both ends of the

- alkene are attacked by an electrophile (E⁺) to form a carbocation intermediate (an alkyl cation).
- The carbocation is then attacked by a nucleophile (Nu⁻) to form the final product, an alkylated alkane.

When reacting alkenes with H⁺, the following mechanism applies:

Electrophilic Addition of H⁺

- In the following mechanism, alkenes react with H⁺ to form a carbocation intermediate. Hydrogen ions are added across the double bond.
- Write down the mechanism for the reaction of propene with H⁺.
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Exercises

- Write down the mechanism for the reaction of propene with H⁺.

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Joacim Rocklöv



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