

## Chemistry I: Periodicity Worksheet

- 1) Arrange the following elements in order of increasing size: Al, Br, C, K, and Na.

100

- 2) Arrange the following elements in order of increasing size: Ca, Rb, P, Ge, and Sr.

100

- 3) Circle the atom or ion in each pair that has the larger radius.

- ✓ **Immer!**

- 4) Select the atom or ion in each pair that has the larger radius.

- c) Error As

- 5) Which of the following groups of elements is arranged correctly in order of increasing ionization energy?  
(Answer by circling the correct answer.)

- d)  $\text{Ne} < \text{C} < \text{Si} < \text{Li}$

- 6) Arrange the following atoms in order of increasing ionization energy: Li, K, C, and N.

1999

- 7) Arrange the following atoms in order of increasing ionization energy: Si, K, As, and Ca.

100

- c) Place the elements in order of increasing ionization energy.

1000

# Periodicity Worksheet Answer Key

**David W. James**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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