

Graphing Periodic Trends Lab

Name _____ Hr _____

PURPOSE: To determine how certain properties are periodic when the elements are arranged in periods or groups.

DISCUSSION: There are many periodic trends that occur as you cross a period or go down a group/family on the Periodic Table. Graphing various elemental properties will help to illustrate these trends.

REFERENCE:

Angstrom - A unit of distance equal to one ten-billionth of a meter.

Atomic Radius - The distance from the center of the nucleus to the valence electron energy level.

Ionization energy - The energy required to remove an electron from an atom.

DATA

Properties of the First 18 Elements

Element	Atomic Number	Atomic Radius (Angstroms)	Ionization Energy
H	1	0.79	314
He	2	0.49	567
Li	3	2.05	124
Be	4	1.4	215
B	5	1.17	191
C	6	0.91	260
N	7	0.75	335
O	8	0.65	314
F	9	0.57	402
Ne	10	?	?
Na	11	2.23	119
Mg	12	1.72	176
Al	13	1.82	138
Si	14	1.46	168
P	15	1.23	242
S	16	1.09	239
Cl	17	0.97	299
Ar	18	?	?

Using the data above, make a line graph the following data. Draw a dark vertical line to separate the 3 periods. Give each graph a title and label both the x-axis and the y-axis. Connect your points. Predict the plot points for elements with question marks.

1. Atomic Radius (y) vs. Atomic Number (x) for the first 3 periods (18 elements – 3 graphs)
2. Ionization energy (y) vs. Atomic Number (x) for the first 3 periods (18 elements – 3 graphs)

Periodic Trends Lab Answers

Jin-Ying Zhang



Periodic Trends Lab Answers:

CBSE Science Chapterwise Case Study Class 10 Priti Singhal, 2024-11-17 This book is structured to align with the latest syllabus and curriculum guidelines ensuring that the content is both relevant and rigorous Each chapter begins with a clear set of learning objectives providing a roadmap for students to understand what they will achieve by the end of the chapter We have included numerous diagrams illustrations and real life examples to make complex concepts more accessible and engaging

Exploring General Chemistry in the Laboratory Colleen F. Craig, Kim N. Gunnerson, 2017-02-01 This laboratory manual is intended for a two semester general chemistry course The procedures are written with the goal of simplifying a complicated and often challenging subject for students by applying concepts to everyday life This lab manual covers topics such as composition of compounds reactivity stoichiometry limiting reactants gas laws calorimetry periodic trends molecular structure spectroscopy kinetics equilibria thermodynamics electrochemistry intermolecular forces solutions and coordination complexes By the end of this course you should have a solid understanding of the basic concepts of chemistry which will give you confidence as you embark on your career in science

Hands-On General Science Activities With Real-Life Applications Pam Walker, Elaine Wood, 2008-04-21 In this second edition of Hands On General Science Activities with Real Life Applications Pam Walker and Elaine Wood have completely revised and updated their must have resource for science teachers of grades 5 12 The book offers a dynamic collection of classroom ready lessons projects and lab activities that encourage students to integrate basic science concepts and skills into everyday life

Using Constructivism in Teaching AP Chemistry Avidah Lotfi, 2004 Developing and Sustaining a Research-supportive Curriculum Kerry K. Karukstis, Timothy E. Elgren, 2007 This compendium of successful curricular and institutional practices to develop critical research skills emphasized the importance of the collective efforts of the undergraduate community to integrate research and education By collecting and disseminating a variety of mechanisms that are effective means of creating a research supportive undergraduate curriculum the Council on Undergraduate Research aims to encourage faculty and institutions to continue to seek creative useful and significant ways to promote learning through research Publisher's description

Laboratory Manual for Principles of General Chemistry Jo Allan Beran, 2010-11-01 This new edition of the Beran lab manual emphasizes chemical principles as well as techniques The manual helps students understand the timing and situations for the various techniques The Beran lab manual has long been a market leading lab manual for general chemistry Each experiment is presented with concise objectives a comprehensive list of techniques and detailed lab intros and step by step procedures

Chemistry in the Laboratory James M. Postma, Julian L. Robert, J. Leland Hollenberg, 2004-03-12 This clearly written class tested manual has long given students hands on experience covering all the essential topics in general chemistry Stand alone experiments provide all the background introduction necessary to work with any general chemistry text This revised edition offers new experiments and expanded information on applications to real world situations

O-level Chemistry

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Chemistry Trends Keith B. Hutton, 2024-12-15 Today's fast moving world of science will have far reaching effects on all of our lives Trends in Science is a series of essential readings for anyone who wants to know more about how his or her future will be affected as well the series provides accessible and stimulating material for high school and college students for researchers and librarians All titles in the series provide an introductory overview of the field in the last 100 years reviewing the past but also predicting the new developments of the future a detailed chronology of the most important milestones an index of key terms and concepts biographies of the most

important scientists in each field and their role in shaping their particular branch of science a listing of important Websites a directory of organizations and suggestions for further reading

Parallel Curriculum Units for Science, Grades 6-12

Jann H. Leppien, Jeanne H. Purcell, 2011-02-15 Teachers at various stages of professional development in curriculum design will find these materials powerful examples that will guide their growth and development and sharpen their skills Mary L. Slade Professor James Madison University Harrisonburg VA Supercharge your science curriculum to challenge all students Based on the best selling book *The Parallel Curriculum* this professional development resource gives multifaceted examples of rigorous learning opportunities for science students in Grades 6-12 The four sample units revolve around genetics the convergence of science and society the integration of language arts and biology and the periodic table The editors and contributors provide user friendly methods for creating more thoughtful lessons and show how to differentiate them for the benefit of all students Included are field tested and standards based strategies that guide students through Exploring the nature of knowledge Discovering connections between science and other subjects Deepening science comprehension according to their interests and abilities Connecting science to society through the study of genetics historic events literature and chemistry Each unit includes subject matter background a content framework study components teacher reflections and sample lessons Also available are online content tools such as handouts PowerPoint presentations and research activities Breathe new life into science learning with this powerful guidebook written by master educators

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Clinical Laboratory Science - E-Book Mary Louise Turgeon, 2015-02-10 Using a discipline by discipline approach Linne perforated pages make it easy for students to remove for use in the lab and for assignment of review questions as homework NEW Instrumentation updates show new technology being used in the lab NEW Additional key terms in each chapter cover need to know terminology NEW Additional tables and figures in each chapter clarify clinical lab science concepts

Contemporary Chemistry Leonard Saland, 1993 This comprehensive guide gives you lesson plans activities and tests for two sequential semester long chemistry courses It is designed to work with our student book *Contemporary Chemistry* Each lesson plan features a DO NOW section to engage students as soon as they get to class instructional objectives an aim for that class period a motivational application questions or demonstrations to help students draw valid conclusions homework assignments You also get term calendars weekly tests and complete answer keys

Carolina Science and Math Carolina

Biological Supply Company, 2003

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bowl whisk together the coconut oil or butter egg whites vanilla extract and liquid stevia add in the greek yogurt stirring

until no large lumps remain stir in $\frac{1}{2}$ cup 2 tablespoons of milk

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