

Observation Table :

Mass of clean, empty crucible and lid (g)	31.072
Mass of crucible, lid, and magnesium (g)	31.638
Mass of crucible, lid, and magnesium oxide (g)	31.992
Mass of magnesium oxide produced (g)	0.906 g
Mass of oxygen that must have reacted (g)	0.336 g
Percent mass of magnesium in magnesium oxide $\frac{m(\text{Mg})}{m(\text{MgO})} \times 100\%$	62.9 % $\approx 63\%$

$$\frac{0.570}{0.906} \times 100 = 62.91$$

Analyze and Interpret

1. What mass of magnesium was used in the reaction?
2. What mass of magnesium oxide was produced?
3. Calculate the mass of oxygen that reacted with the magnesium.

Conclude and Communicate

4. Use the data to calculate the percentage composition of magnesium oxide.
5. Calculate the percentage error (PE) using the difference between the experimental mass percent (EP) of magnesium and the actual mass percent (AP) of magnesium.
6. Suppose that some magnesium oxide smoke escaped during the investigation. Would the Mg : O ratio have increased, decreased, or remained unchanged? Explain your answer using sample calculations.
7. How would the value you calculated for the percentage composition of magnesium oxide have been affected if all the magnesium in the crucible had not reacted? Support your answer with sample calculations.

Percentage Composition Of Magnesium Oxide Lab Answers

D Keegan



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Report of Investigations ,1961 **Merrill Chemistry** Robert C. Smoot,Smoot,Richard G. Smith,Jack Price,1998
The Elements of Life Alan Sherman,Sharon J. Sherman,1979-12 **Concrete-cement Age** ,1915 *Concrete-cement Age* Allen Brett,Harvey Whipple,1915 **Report summaries** United States. Environmental Protection Agency,1983
Scientific and Technical Aerospace Reports ,1991-03 *Report of the Division of Chemistry* Canada. Department of Agriculture. Division of Chemistry,1924 **Nuclear Science Abstracts** ,1975 **Report of the State Board of Geological Survey of Michigan for the Year ...** Michigan. Geological Survey,1903 **Report of the Division of Chemistry** Canada. Experimental farms. Division of Chemistry,1920 **Energy Research Abstracts** ,1990 *Journal of Scientific & Industrial Research* ,1954 **U.S. Government Research Reports** ,1964 Proceedings, Environmental Research - Technology Transfer Conference ,1993 **U.S. Government Research and Development Reports** ,1965
Report of the Department of Mines for the Year ... Western Australia. Department of Mines,1949 **Holt Chemistry** Ralph Thomas Myers,2004 *Annual Progress Report* Geological Survey of Western Australia,1924 *ERDA Energy Research Abstracts* ,1977

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