

# OPTICS

## Fourth Edition

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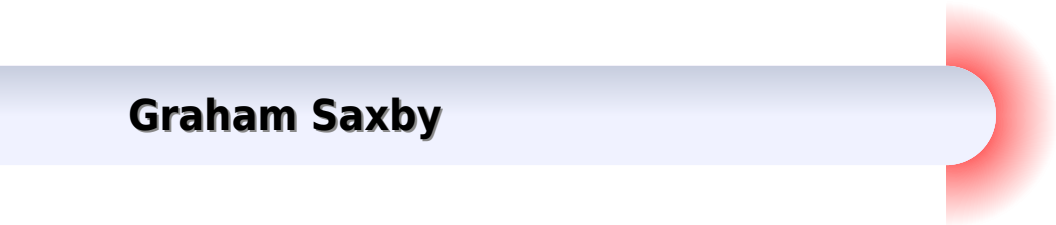
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# Optics Eugene Solutions Manual

**Graham Saxby**



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web chemlab 16 calorimetry answers 1 the reaction you will be determining the  $\Delta H$  is the reaction with  $\text{NH}_3(\text{aq})$  with  $\text{HCl}(\text{aq})$  write the balanced equation for this reaction  $\text{NH}_3(\text{aq}) + \text{HCl}(\text{aq}) \rightarrow \text{NH}_4\text{Cl}(\text{aq})$  2 in addition you will be verifying this by doing two other reactions which add up to the reaction above write balanced equations for these

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