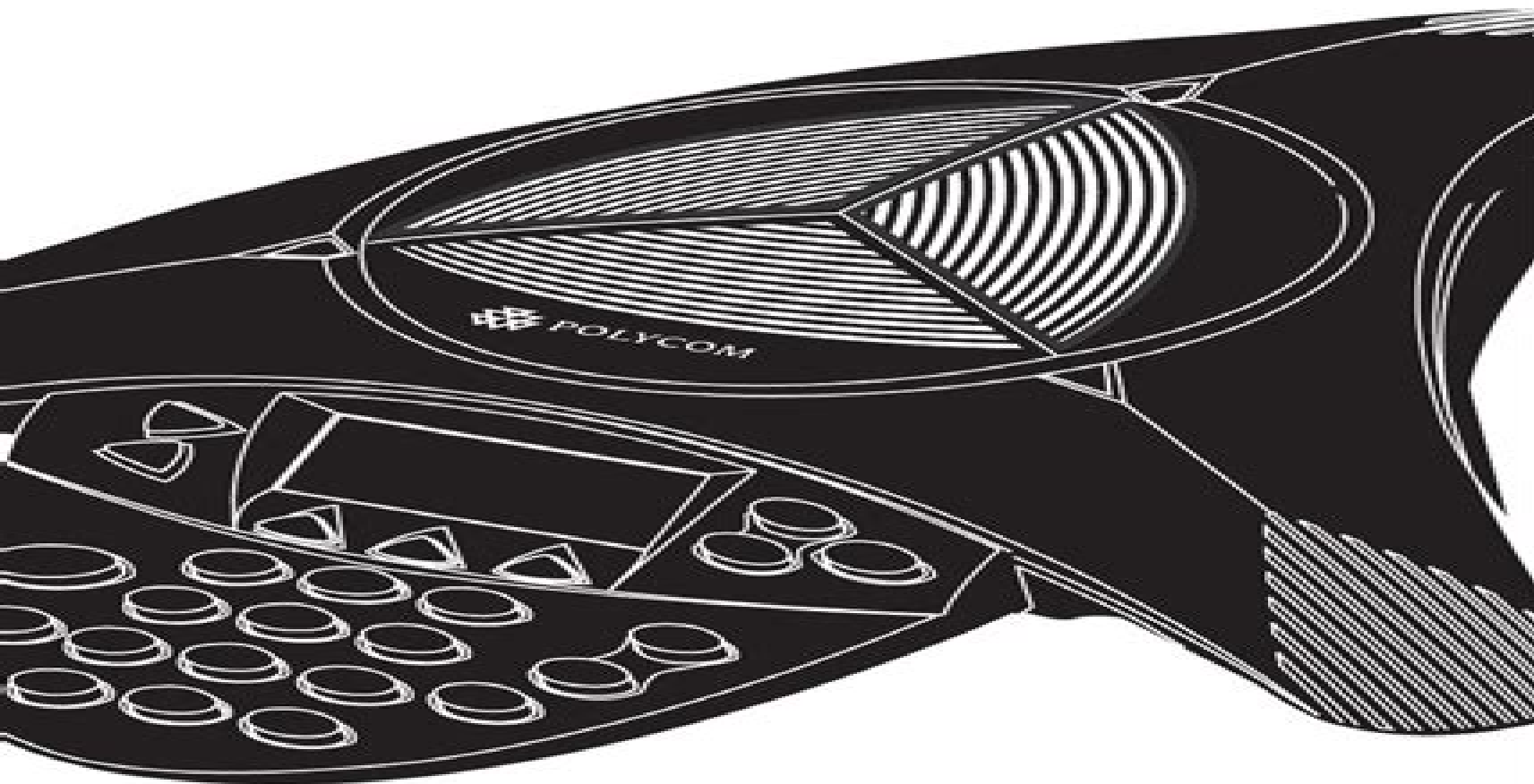


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web as a result of the reaction the equilibrium concentration of $NH_3(g)$ became 0.540 M what is the value of the equilibrium constant for this reaction at the given temperature of 700°C $N_2(g) + 3H_2(g) \rightleftharpoons 2NH_3(g)$ i write the initial concentrations of reactants and products

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