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# CLASSIFYING CHEMICAL REACTIONS

Classify the following chemical reactions as one of the following: synthesis, decomposition, single-replacement, double-replacement, or combustion.

$2H_0 + O_0 \rightarrow 2H_0O$	Caco, Cao + Co,	
3K + 3H <sub>2</sub> O → H <sub>4</sub> + 3KOH	CHOO, • 2HCI CHOL • HLCO.	
290010 <sub>9</sub> 28001 + 30 <sub>9</sub>	Ca + 2HCl CaCl <sub>2</sub> + H <sub>2</sub>	
280 <sub>e</sub> + 0 <sub>e</sub> 280 <sub>e</sub>	PERNOJ, - SHCI - PECL - SHEED,	
$Ov + 2AgNO_y \rightarrow 2Ag + OutyOy$	2Na • Cl <sub>e</sub> 2NeCli	
2H <sub>2</sub> O 2H <sub>2</sub> - O <sub>2</sub>	CH <sub>6</sub> + 20 <sub>5</sub> → CO <sub>3</sub> + 24 <sub>5</sub> O	
$PbONO_{\mathcal{O}_{S}} = 290  \rightarrow  PbI_{b} = 29000_{b}$	H,CO, CO, - H,O	
2Fe + O₂ → 2FeO	Zn + 2HCt H <sub>b</sub> + ZnCt <sub>b</sub>	
HCI + NaOH → NaOI + HOH	25% + 254,0 25%0H + H <sub>2</sub>	

# **Pre Lab Answers To Classifying Chemical Reactions**

**Wolfgang Guggemos** 

**Pre Lab Answers To Classifying Chemical Reactions:** 

Unveiling the Magic of Words: A Report on "Pre Lab Answers To Classifying Chemical Reactions"

In some sort of defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is actually aweinspiring. Enter the realm of "**Pre Lab Answers To Classifying Chemical Reactions**," a mesmerizing literary masterpiece penned by way of a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book is central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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