# Pre- AP Chemistry ~ Unit 6 - Notes Packet - chemical reactions

- Chemical Reaction Definition: the process by which \_\_one or more substances are changed\_\_ into one or more different substances
- How do we know a chemical change has occurred?
  - Evidence that a chemical change may have occurred can include:
  - Production of <u>bubbles</u> (gas)
  - Formation of a \_\_precipitate\_\_\_(solid)
  - Production of \_\_\_\_heat and light
  - Color change
  - Noise
  - Ultimately, only the formation of a \_\_NEW\_\_\_ substance can be absolute proof that a chemical change has occurred.
- How do we write chemical reactions?
  - Include multiple molecules
  - Reactants ----> products
  - (original substances) (resulting substances)
- How Do Reactants Turn Into Products?
  - Electrons from atoms want to get into the lowest energy state possible
  - They will rearrange themselves with other atoms to do this
  - Reactants come into contact and collide
  - Bonds between reactant atoms break.
  - This gives freedom to atoms to rearrange
  - Atoms form bonds with new atoms to create products
- Reactions and Energy
  - One of the keys to making reactions "go" is \_\_energy\_\_\_\_\_\_:
  - Energy is the ability to do work (to move or change matter)
  - Most reactions will either \_\_release or absorb \_\_\_ energy

# **Pre Ap Chemistry Unit 6 Hw Packet Key**

**E Durkheim** 

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