

# Reference Tables for Physical Setting/EARTH SCIENCE

## Radioactive Decay Data

| RADIOACTIVE ISOTOPE | DISINTEGRATION                                                                           | HALF-LIFE (years)    |
|---------------------|------------------------------------------------------------------------------------------|----------------------|
| Carbon-14           | $^{14}\text{C} \rightarrow ^{14}\text{N}$                                                | $5.7 \times 10^3$    |
| Potassium-40        | $^{40}\text{K} \rightarrow ^{40}\text{Ar}$<br>$^{40}\text{K} \rightarrow ^{40}\text{Ca}$ | $1.3 \times 10^9$    |
| Uranium-238         | $^{238}\text{U} \rightarrow ^{206}\text{Pb}$                                             | $4.5 \times 10^9$    |
| Rubidium-87         | $^{87}\text{Rb} \rightarrow ^{87}\text{Sr}$                                              | $4.9 \times 10^{10}$ |

## Specific Heats of Common Materials

| MATERIAL          | SPECIFIC HEAT (Joules/gram • °C) |
|-------------------|----------------------------------|
| Liquid water      | 4.18                             |
| Solid water (ice) | 2.11                             |
| Water vapor       | 2.00                             |
| Dry air           | 1.01                             |
| Basalt            | 0.84                             |
| Granite           | 0.79                             |
| Iron              | 0.45                             |
| Copper            | 0.38                             |
| Lead              | 0.13                             |

## Equations

$$\text{Eccentricity} = \frac{\text{distance between foci}}{\text{length of major axis}}$$

$$\text{Gradient} = \frac{\text{change in field value}}{\text{distance}}$$

$$\text{Rate of change} = \frac{\text{change in value}}{\text{time}}$$

$$\text{Density} = \frac{\text{mass}}{\text{volume}}$$

## Properties of Water

|                                          |          |
|------------------------------------------|----------|
| Heat energy gained during melting        | 334 J/g  |
| Heat energy released during freezing     | 334 J/g  |
| Heat energy gained during vaporization   | 2260 J/g |
| Heat energy released during condensation | 2260 J/g |
| Density at 3.98°C                        | 1.0 g/mL |

## Average Chemical Composition of Earth's Crust, Hydrosphere, and Troposphere

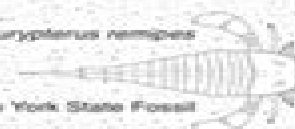
| ELEMENT (symbol) | CRUST           |                   | HYDROSPHERE       | TROPOSPHERE       |
|------------------|-----------------|-------------------|-------------------|-------------------|
|                  | Percent by mass | Percent by volume | Percent by volume | Percent by volume |
| Oxygen (O)       | 46.10           | 94.04             | 33.0              | 21.0              |
| Silicon (Si)     | 28.20           | 0.88              |                   |                   |
| Aluminum (Al)    | 8.23            | 0.48              |                   |                   |
| Iron (Fe)        | 5.63            | 0.49              |                   |                   |
| Calcium (Ca)     | 4.15            | 1.18              |                   |                   |
| Sodium (Na)      | 2.36            | 1.11              |                   |                   |
| Magnesium (Mg)   | 2.33            | 0.33              |                   |                   |
| Potassium (K)    | 2.09            | 1.42              |                   |                   |
| Nitrogen (N)     |                 |                   |                   | 78.0              |
| Hydrogen (H)     |                 |                   | 66.0              |                   |
| Other            | 0.91            | 0.07              | 1.0               | 1.0               |

### 2011 EDITION

This edition of the Earth Science Reference Tables should be used in the classroom beginning in the 2011–12 school year. The first examination for which these tables will be used is the January 2012 Regents Examination in Physical Setting/Earth Science.

*Eurypterus remipes*

New York State Fossil



# Practice Earth Science Reference Table Questions

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