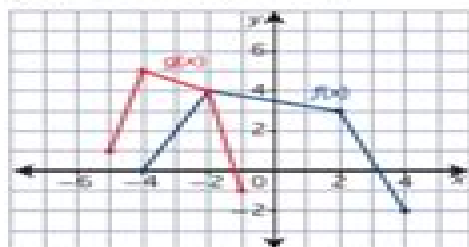


Unit 1 Test

Multiple Choice

For #1 to #7, choose the best answer.

1. The graph of $f(x)$ and its transformation, $g(x)$, are shown below.



The equation of the transformed function is

- A $g(x) = f\left(\frac{1}{2}(x - 3)\right) + 1$
 B $g(x) = f(2(x - 3)) + 1$
 C $g(x) = f\left(\frac{1}{2}(x + 3)\right) + 1$
 D $g(x) = f(2(x + 3)) + 1$
2. The graph of the function $y = f(x)$ is transformed by a reflection in the y -axis and a horizontal stretch about the y -axis by a factor of 3. Which of the following will not change?
- I the domain
 II the range
 III the x -intercepts
 IV the y -intercept
- A I only
 B I and III
 C II and IV
 D depends on $y = f(x)$

3. Which pair of functions are *not* inverses of each other?

- A $f(x) = 5x$ and $g(x) = \frac{x}{5}$
 B $f(x) = x + 3$ and $g(x) = x - 3$
 C $f(x) = 4x - 1$ and $g(x) = \frac{1}{4}x + \frac{1}{4}$
 D $f(x) = \frac{x}{2} + 5$ and $g(x) = 2x - 5$

4. Which function has a domain of $\{x \mid x \in \mathbb{R}\}$ and a range of $\{y \mid y \geq -3, y \in \mathbb{R}\}$?

- A $y = |x + 4| - 3$
 B $y = \sqrt{x + 4} - 3$
 C $y = \sqrt{x^2 - 4} - 3$
 D $y = (x - 4)^2 - 3$

5. If the graph of $y = \sqrt{x + 3}$ is reflected in the line $y = x$, then which statement is true?

- A All invariant points lie on the y -axis.
 B The new graph is not a function.
 C The point $(6, 3)$ will become $(-3, 6)$.
 D The domain of the new graph is $\{x \mid x \geq 0, x \in \mathbb{R}\}$.

6. If the graph of a polynomial function of degree 3 passes through $(2, 4)$ and has x -intercepts of -2 and 3 only, the function could be

- A $f(x) = x^3 + x^2 - 6x - 12$
 B $f(x) = x^3 - x^2 - 6x + 12$
 C $f(x) = x^3 - 4x^2 - 3x + 18$
 D $f(x) = x^3 + 4x^2 - 3x - 18$

7. If $P(x) = -x^3 - 4x^2 + x + 4$, then

- A $x + 1$ is a factor
 B $P(0) = -1$
 C the y -intercept is -4
 D $x - 1$ is not a factor

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web science chemistry chemistry questions and answers ch 201 lab exercise molecular modeling purpose to practice lewis dot structures and associate them with models to use molecular modeling software and compare the results to static models discussion in this exercise you will get practice drawing electron dot structures for organic compounds

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web 2019 0501140012 american native medicine 1 2019 0612171451 orpic company 1 biome viewer student ws copy of lab 1 effect environment se the periodic properties of the main group to understand theory behind lewis structure and valence shell electron pair repulsion theory to experience molecule s geometrical shapes by the model that has

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web mar 27 2023 a the lewis structure for water is represented by the given option a in this structure the oxygen atom is in the center and is bonded to two hydrogen atoms the oxygen atom has two lone pairs of electrons this structure follows the octet rule where each atom has a full valence shell with eight electrons

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web introduction the physical and chemical behavior of compounds depends on the types of bonds that exist within the molecule and on the overall shape of the molecules the branch of chemistry that investigates how molecules are put together is

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web lewis structures represent the valence electrons present in the molecule as lone pairs unshared electrons drawn as dots or bonds shared electrons drawn as lines procedure for drawing lewis structures add up the total number of valence electrons that each atom contributes to the molecule polyatomic ion

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web the pcl 3 lewis structure has the typical case of phosphorus p in the center with 3 bonds to 3 other atoms phosphorus is from the same column as nitrogen in the periodic table meaning that p and n generally have the same bonding structure note the lone pair dots without bonds on top of p just like for n in the previous example for nh 3 chlorine cl

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web step 1 figure out how many electrons the molecule must have based on the number of valence electrons in each atom when drawing the structure of an ion be sure to add subtract electrons to account for the charge step 2 connect the atoms to each other with single bonds to form a skeleton structure

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web molecular models and 3d printing activity lewis dot structures and molecule geometries worksheet answer key 6 10 co 3 2 total number of valence electrons 24 cad engineered 3d sketch model show dipole arrows lewis structure is there a polar bond in this molecule yes or no vsepr shape name trigonal planar bond angles 120

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web jun 8 2022 lewis structure in chemistry lewis structures also called electron dot structures or electron dot diagrams are diagrams that show the bonding between atoms of a molecule and the lone pairs of

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web this exercise required us to take the formulas in the data table and calculate the number of valence electrons using the periodic table draw its lewis structure on paper build the molecular model and record a photo list the number of bonds and lone pairs surrounding the central atom and lastly identify the structure geometry