

Use the following information to answer questions 12-16.

trial	initial [A]	initial [B]	initial [C]	rate
#1	0.10	0.02	0.04	10 M/hr
#2	0.10	0.03	0.04	15 M/hr
#3	0.20	0.02	0.08	80 M/hr
#4	0.20	0.02	0.16	160 M/hr
#5	0.05	0.01	0.08	?

$$\Delta []^x = \Delta \text{Rate}$$

$$\left(\frac{1}{2}\right)^2 \left(\frac{1}{2}\right) \left(2\right) = \frac{1}{4} 10 \text{ M/hr}$$

12. What is the rate order for reactant C?

1st order

13. What is the rate order for reactant B?

1st order

14. What is the rate order for reactant A? (Hint: Use the information for the question 12 and trials #1 and #3)

2nd order

15. Calculate the rate constant and write out a rate law equation for the data given above.

$$\text{Rate} = k [A]^2 [B]^1 [C]^1$$

16. What would the rate be in trial #5?

2.5 M/hr Rate overall = 4

$$k = \frac{\text{Rate}}{[(A)^2 (B)^1 (C)^1]} \quad \frac{1}{M^3 s}$$

Practice Rate Law Problems Answers

Sebastian Brünink



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Survival Guide to General Chemistry Patrick E. McMahon, Rosemary McMahon, Bohdan Khomtchouk, 2019-02-13 This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics The focus is not to recap or review the theoretical concepts well described in the available texts Instead the topics and descriptions in this book make available specific detailed step by step methods and procedures for solving the major types of problems in general chemistry Explanations instructional process sequences solved examples and completely solved practice problems are greatly expanded containing significantly more detail than can usually be devoted to in a comprehensive text Many chapters also provide alternative viewpoints as an aid to understanding Key Features The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester Each is written in a specific and detailed step by step process for problem solving whether mathematical or conceptual Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts Includes a chapter designed to eliminate confusion concerning acid base reactions which often persists through working with acid base equilibrium Many chapters provide alternative viewpoints as an aid to understanding This book addresses a very real need for a large number of incoming freshman in STEM fields **(Free Sample) GO TO Objective NEET Chemistry Guide with DPP & CPP**

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written in a conversational question based format with a well defined problem solving strategy and presented in a way to encourage readers to think like a chemist and to think outside of the box Numerous examples are presented in every chapter to aid students and provide helpful self learning tools The topics are arranged throughout the textbook in a traditional approach to the subject with the primary audience being undergraduate students and advanced high school students of chemistry

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