

Use the partial periodic table to answer questions 4 -7.

⁶
C
Carbon
12.0107

— Atomic number
 — Symbol
 — Name
 — Average Atomic Mass

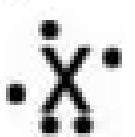
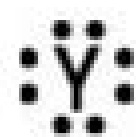
Partial Periodic Table of the Elements

IA 1	H Hydrogen 1.00794	IIA 2	He Helium 4.0026
2	Li Lithium 6.941	Be Beryllium 9.0122	
3	Na Sodium 22.989769	Mg Magnesium 24.3050	
4	K Potassium 39.0983	Ca Calcium 40.078	

IIIA 13	IVA 14	VA 15	VIA 16	VIIA 17	VIIIA 18
B Boron 10.811	C Carbon 12.0107	N Nitrogen 14.0064	O Oxygen 15.9994	F Fluorine 18.9984	Ne Neon 20.1797
Al Aluminum 26.981538	Si Silicon 28.0855	P Phosphorus 30.97376	S Sulfur 32.06	Cl Chlorine 35.453	Ar Argon 39.948

Question 5

Suppose scientists discovered four new elements (W, X, Y, Z) while studying rock and soil samples brought back from a Mars mission. Which Lewis dot structure represents an element that should be placed in column VIIA (17) of the periodic table?

- A. 
- B. 
- C. 
- D. 

Partial Periodic Answer Physical Science Lab

Siddappa N.Byrareddy



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