

Practice Final Exam, Chemistry 2220, Organic Chem II

1. Rank the designated protons by chemical shift ( $\delta$ ), highest  $\delta$  first.



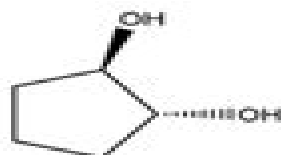
- A. IV > III > II > I    B. IV > II > III > I    C. III > IV > II > I    D. III > IV > I > II

2. What are the products, if any, expected from the following reaction?



- $\text{CH}_3\text{OCH}_3 + \text{NaHS}$      $\text{CH}_3\text{SCH}_3 + \text{NaOH}$      $\text{CH}_3\text{O}-\text{H} + \text{CH}_3\text{S}^-\text{Na}^+$     no reaction  
 A.                                      B.                                      C.                                      D.

3. What is the IUPAC name for the following compound?



- A. cis-1,2-cyclopentanediol  
 B. meso-1,2-cyclopentanediol  
 C. (1R,2R)-1,2-cyclopentanediol  
 D. (1R,2S)-1,2-cyclopentanediol

4. Which of the following explains why cyclohexanol has a  $\text{pK}_a$  of 18 and phenol has a  $\text{pK}_a$  of 10?

- A. Phenolate is a stronger base than the conjugate base of cyclohexanol  
 B. The conjugate base of cyclohexanol is resonance stabilized  
 C. The conjugate base of phenol is resonance stabilized  
 D. Phenol is a weaker acid than cyclohexanol

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