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acid the species produced when a base accepts a hydrogen ion from an acid

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arrhenius model of bases a substance that contains a hydroxide group and dissociates to produce a hydroxide

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web kw the product of the concentrations of the hydrogen ions and hydroxide ions acidic solution one which h is greater than

the oh basic solution one in which the h is less than the oh alkaline solutions the h of a basic solution is less than 1×10^{-7} m

they are the same as basic solutions

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molecule that gains a hydrogen ion becomes this which is positively charged what did lewis propose an acid accepts a pair of

electrons during a reaction while a base donates a pair of electrons

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