Polyatomic Ions Worksheet Answer Key

Polyatomic Practice

1. Name or write the formula for the following polyatomic ions

sulfate	SO?	CO ₃ 2-	carbonate	
nitrite	NO;	MnO ₃ 1-	manganate	
perphosphate	PO3-	SO ₅ ²⁻	persulfate	
hypoiodite	10	BrO ₂ 1-	bromite	
chlorite	CIO,	OO42-	percarbonate	
phosphite	PO3-	POs ³⁻	perphosphate	
percarbonate	CO ₄	CIO ¹⁻	hypochlorite	
bromate	BrO;	1O25-	iodite	
hyposulfite	SO ₂ -	PO43-	phosphate	
permanganate	MnO.	NO ₂ 1-	nitrite	
carbonite	CO2	SO ₄ 2-	sulfate	

2. Name or write the formula for the following Type I polyatomic ionic compounds

beryllium hydroxide	Be(OH) ₂	Ba(IO ₃) ₂	barium iodate
sodium nitrite	NaNO ₂	Ga(CNO) ₃	gallium cyanate
ammonium chloride	NH,CI	Ag ₂ SO ₃	silver sulfite
calcium bisulfate	Ca(HSO ₄) ₂	MgCO	magnesium hypocarbonite
rubidium perchlorate	RECIO.	NH ₃ NO ₂	ammonium nitrite
strontium sulfite	SrSO,	$AI(C_2H_3O_2)_3$	aluminum acetate
aluminum acetate	AI(C ₂ H ₃ O ₂) ₃	SrSO ₅	strontium persulfate
ammonium nitrate	NH,NO,	RbClO ₂	rubidium chlorite
magnesium hypocarbonite	MgCO	Ca(HSO ₄) ₂	calcium bisulfate
silver hyposulfite	Ag ₂ SO ₂	NH ₄ CI	ammonium chloride

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S Marginson

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