#### 1. (4 points)

The specific heat (s) of iron (Fe) is 0.40 J/(g K). Calculate the heat effect q when 20 g of iron is heated from 100 °C to 200 °C. Include the sign!

 $q = m.s.\Delta T$  for  $\Delta T \Delta K = \Delta^{\circ}C$  q = 20x0.40x(200-100) = +800 J

You must include the units and sign in the answer!

#### 2. (4 points)

Reaction:  $Mg(s) + \frac{1}{2}O_3(g) \rightarrow MgO(s)$   $\Delta H = -600 \text{ kJ/mol}$ 

The Molar Mass of Mg = 24.0 g/mol.

Calculate the enthalpy change when 2.40 g Mg(s) reacts with excess oxygen. Include the sign!

 $2.40 \text{ g Mg} = 2.40/24.0 = 0.1 \text{ mol} \rightarrow \Delta H = 0.1(\text{mol})x(-600)(kJ/\text{mol}) = -60 \text{ kJ}$ 

Again, your answer must include both the sign and the units.

 (2 points) Which one of the following four reactions is a formation reaction? <u>Circle</u> the correct answer.

a) 
$$H_2O(I) + \frac{1}{2}O_2 \longrightarrow H_2O_2(I)$$

b) 
$$SO_2(g) + \frac{1}{2}O_2 \rightarrow SO_2(g)$$

c) 
$$Cl(g) + H(g) \rightarrow HCl(g)$$

d) 
$$S(s) + O_2(g) \rightarrow SO_2(g)$$

Look up the definition of a formation reaction in Chang or in your lecture notes: the compound is formed from its elements, and the elements must be as they naturally occur, so e.g.  $Cl_2$  not  $Cl_4$  and  $H_2$  not  $H_3$ !

#### CHEM 101 Section L52 Quiz #6 b

#### 1. (4 points)

The specific heat (s) of copper (Cu) is 0.38 J/(g.K). Calculate the heat effect q when 10 g of copper is heated from 100 °C to 300 °C. Include the sign!

 $q = m.s.\Delta T$  for  $\Delta T \Delta K = \Delta^*C$  q = 10x0.38x(300-100) = +760 J

You must include the units and sign in the answer!

#### 2. (4 points)

Reaction:  $Ca(s) + \frac{1}{2}O_2(g) \rightarrow CaO(s)$ 

$$\Delta H = -650 \text{ kJ/mol}$$

The Molar Mass of Ca = 40.0 g/mol.

Calculate the enthalpy change  $\Delta H$  when 8.0 g Ca(s) reacts with excess oxygen. Include the sign!!

 $8.0 \text{ g Ca} = 8.0/40/ = 0.2 \text{ mol} \rightarrow \Delta H = 0.2(\text{mol})x(-650)(kJ/\text{mol}) = -130 \text{ kJ}$ 

Again, your answer must include both the sign and the units.

 (2 points) Which one of the following reactions is a formation reaction? <u>Circle</u> the correct answer.

a) 
$$Br(g) + H(g) \rightarrow HBr(g)$$

b) 
$$H_2O(I) + \frac{1}{2}O_2 \rightarrow H_2O_2(I)$$

c) 
$$2P(s) + 2\%O_2(g) \rightarrow P_2O_3(s)$$

d) 
$$SO_2(g) + \frac{1}{2}O_2 \rightarrow SO_3(g)$$

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**G** Thomas

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practitioners administrators policymakers and advocates how and why this particular reform of conditions achieved successful outcomes when others failed Maintaining that the Chicago Crime Lab findings are the gold standard evidence based research EBR in pretrial detention Roush holds that the observed firsts for juvenile detention may perhaps have the power to transform all custody practices He shows that the findings validate a new model of institutional reform based on cognitive behavioral programming CBT reveal statistically significant reductions in in custody violence and recidivism and demonstrate that at least one variation of short term secure custody can influence positively certain life outcomes for Chicago s highest risk and most disadvantaged youth With the Quarterly Journal of Economics imprimatur and endorsement by the President's Council of Economic Advisors the book is a reverse engineering of these once in a lifetime events recidivism reduction and EBR in pretrial detention that explains the important and transformative implications for the future of juvenile justice practice The book is essential reading for graduate students in juvenile justice criminology and corrections as well as practitioners judges and policymakers Chemical Engineering and Mining Review ,1927 Minor Minerals. Major Implications: Using Key Mineral Phases to Unravel the Formation and Evolution of Earth's Crust V. van Schijnde, K. Cutts, I. Pereira, M. Guitreau, S. Volante, M. Tedeschi, 2024-01-11 The investigation of key mineral phases such as zircon apatite titanite rutile monazite xenotime allanite baddeleyite and garnet explored in this book has provided breakthroughs in our understanding of continental crust composition and evolution as well as the timing conditions petrogenetic and geodynamic processes related to its growth and reworking Therefore the continuing development of analytical techniques improvement of tools data handling processing and interpretation allow us to extract and better understand these complex geological processes This special publication aims at showcasing contributions reviewing the tools and applications of these key minerals recent technique developments and new applications using focused case studies investigating igneous metamorphic and or detrital rocks that help us put together the continental crust evolution puzzle This volume highlights the progress made in studies using these key minerals and their future potential Directory of **Distance Learning Opportunities** Modoc Press, Inc., 2003-02-28 This book provides an overview of current K 12 courses and programs offered in the United States as correspondence study or via such electronic delivery systems as satellite cable or the Internet The Directory includes over 6 000 courses offered by 154 institutions or distance learning consortium members Following an introduction that describes existing practices and delivery methods the Directory offers three indexes Subject Index of Courses Offered by Level Course Level Index Geographic Index All information was supplied by the institutions Entries include current contact information a description of the institution and the courses offered grade level and admission information tuition and fee information enrollment periods delivery information equipment requirements credit and grading information library services and accreditation **Christian Home Educators' Curriculum Manual** Cathy Duffy, 1995 The premiere guide for choosing homeschool curriculum For beginners or veterans Cathy helps you wade

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