

Periodic Table WebQuest

Periodic Table of the Elements

Lanthanide Series

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|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|
| La | Ce | Pr | Nd | Pm | Sm | Eu | Gd | Tb | Dy | Ho | Er | Tm | Yb | Lu |
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Actinide Series

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|----|----|----|---|----|----|----|----|----|----|----|----|----|----|----|
| Ac | Th | Pa | U | Np | Pu | Am | Cm | Bk | Cf | Es | Fm | Md | No | Lr |
|----|----|----|---|----|----|----|----|----|----|----|----|----|----|----|

Directions: In each of the websites listed, find the information asked for.

Use <http://atomictimeline.net/index.php> to write the contributions of each person toward the development of their theory of the atom (atomic model) and the Periodic Table of Elements.

1. Democritus
2. John Dalton
3. J. J. Thompson
4. Niels Bohr
5. Ernest Rutherford

Use http://www.kyrene.org/Staff/mijohns/atomic_structure_website/electroncloud.htm, http://education.jlab.org/qa/particlemass_02.html, and <http://www.colorado.edu/physics/2000/isotopes/index.html>, to answer the following:

6. List 3 unique aspects of the electron cloud model in general comparison to previous models.
7. How does the mass of a proton compare to the mass of a neutron?
8. How does the mass of a proton compare to the mass of an electron?

Periodicity Webquest Answer Key

Rachel S Tattersall



Periodicity Webquest Answer Key:

Fuel your quest for knowledge with Learn from is thought-provoking masterpiece, Explore **Periodicity Webquest Answer Key** . This educational ebook, conveniently sized in PDF (PDF Size: *), is a gateway to personal growth and intellectual stimulation. Immerse yourself in the enriching content curated to cater to every eager mind. Download now and embark on a learning journey that promises to expand your horizons. .

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