5.1 Atomic structure and the Periodic Table

year weill learn:

- To understand how the position of an element in the Periodic Table can be used to predict its atomic structure and chemical and physical properties
- To understand that models and analogies can change according to scientific evidence
- · To describe the uses, strengths and limitations of models and analogies
- To represent scientific ideas using recognised symbols or formulae

| | Meaning |
|---|---|
| Electrons | Particles with no charge found in an atom |
| Neutrons | The central part of the atom |
| Protons | The paths or orbits that electrons move along in the atom |
| Nucleus | Very small negatively charged particles in an atom |
| Shells | Particles with a positive charge found in an atom |
| Complete the s | |
| Complete the s Use the words atomic | |

Periodic Table Workbook Answers Chemistry

Ying Liu

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