Chemical bonding



Name _____ Class ____ Date ____

How many atoms of oxygen (O) are in the chemical formula below?

A4 Al₂(SO₄)₃

The diagram shown below is a structural formula for the gas propane. What would be the correct chemical formula for propane?

3 Carbon (C) and hydrogen (H) create covalent bonds to share electrons between them. What is the total number of electrons being shared between carbon and hydrogen in the molecule

A 8 C 16 H-C-C-C-I B 11 D 32 H H H

- The reason why the bonds between carbon (C) and hydrogen (H) H H H H Are covalent is H C C H
 - electrons to give away

 C and H are both nonmetals

 C C and H are both metalloids

 C and H can bond only
 by sharing electrons

The diagram below represents the structural formula for several water molecules. How many molecules are represented?

A 3 B 5 C 10 D 15 6 Explain the arrangement of the molecules in the diagram of water below.

A it is a lucky arrangement

B water molecules are sticky

C water molecules have

D water molecules are magnetic

Water molecules have some polarity in their structure. Using the diagram below, notice how the ends of H atoms (positive) and O atoms (negative) meet. This property is the reason that

A water forms droplets easily
B water boils easily
C ice melts easily
D water has color

Since H atoms are slightly positive and the O atoms are slightly negative, molecules of water have polarity. This makes water a good

A solute
B refrigerant
C solvent
D magnet

The chemical formula for the exygen that we breathe is

B 0, C 0, D 20



How many molecules of oxygen are represented in the chemical formula below?

> A 2 B 4 C 6 D 18

40

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- 1. Understanding the eBook Oaa Practice Science 8th Grade
 - The Rise of Digital Reading Oaa Practice Science 8th Grade
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Oaa Practice Science 8th Grade
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Oaa Practice Science 8th Grade
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Oaa Practice Science 8th Grade
 - Personalized Recommendations
 - Oaa Practice Science 8th Grade User Reviews and Ratings
 - Oaa Practice Science 8th Grade and Bestseller Lists
- 5. Accessing Oaa Practice Science 8th Grade Free and Paid eBooks
 - o Oaa Practice Science 8th Grade Public Domain eBooks

- Oaa Practice Science 8th Grade eBook Subscription Services
- Oaa Practice Science 8th Grade Budget-Friendly Options
- 6. Navigating Oaa Practice Science 8th Grade eBook Formats
 - ∘ ePub, PDF, MOBI, and More
 - Oaa Practice Science 8th Grade Compatibility with Devices
 - Oaa Practice Science 8th Grade Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - o Adjustable Fonts and Text Sizes of Oaa Practice Science 8th Grade
 - Highlighting and Note-Taking Oaa Practice Science 8th Grade
 - Interactive Elements Oaa Practice Science 8th Grade
- 8. Staying Engaged with Oaa Practice Science 8th Grade
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Oaa Practice Science 8th Grade
- 9. Balancing eBooks and Physical Books Oaa Practice Science 8th Grade
 - Benefits of a Digital Library
 - o Creating a Diverse Reading Collection Oaa Practice Science 8th Grade
- 10. Overcoming Reading Challenges
 - o Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Oaa Practice Science 8th Grade
 - Setting Reading Goals Oaa Practice Science 8th Grade
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Oaa Practice Science 8th Grade
 - Fact-Checking eBook Content of Oaa Practice Science 8th Grade
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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