

Stoichiometry Review: Test Review



1. What is stoichiometry? [] Study of quantitative relationships in chemical reaction.
2. On which law is stoichiometry based? [] Law of conservation of mass.
3. According to the law of conservation of mass, the mass of the [] ^{reactants} must [equal] the mass of the ^{products}.
4. What is a mole ratio? [] fractions/ratio comparing two substances.
5. Given the following balanced equation: $2\text{Al} + 3\text{Br}_2 \rightarrow 2\text{AlBr}_3$, determine the mole ratio used for calculating moles of aluminum bromide from moles of bromine. [] $2\text{mol AlBr}_3 / 3\text{mol Br}_2$
6. What must you have before beginning any stoichiometric problem? [] balanced equation
7. What is a limiting reactant? [] runs out
8. What is an excess reactant? [] left over
9. A booklet is prepared from 2 covers, 3 staples, and 20 pages. Assume we have 60 covers, 120 staples, and 400 pages.
 - a. What is the largest number of complete booklets that can be made? [20]
 - b. Which is the "limiting reactant" in this system? [pages]
10. How many moles of lithium chloride will be formed by the reaction of chlorine with 0.046 mol of lithium bromide in the following reaction?
 $[2] \text{LiBr} + [1] \text{Cl}_2 \rightarrow [2] \text{LiCl} + [1] \text{Br}_2$

$$\frac{0.046 \text{ mol LiBr}}{2 \text{ mol LiBr}} \times \frac{2 \text{ mol LiCl}}{2 \text{ mol LiBr}} = 0.046 \text{ mol LiCl}$$

11. Hydrogen peroxide breaks down, releasing oxygen, in the following reaction. What mass of oxygen is produced when 1.840 mol of H_2O_2 decomposes?

$$\frac{1.840 \text{ mol H}_2\text{O}_2}{2 \text{ mol H}_2\text{O}_2} \times \frac{1 \text{ mol O}_2}{2 \text{ mol H}_2\text{O}_2} \times \frac{32.09 \text{ g O}_2}{1 \text{ mol O}_2} = 29.44 \text{ g O}_2$$

12. Hydrogen is generated by passing hot steam over iron, which oxidizes to form Fe_2O_3 , in the following equation. If 625 g of Fe_2O_3 is produced in the reaction, how many moles of hydrogen are produced at the same time?

Review Stoichiometry Answer Key Mixed Review

Richard Bailey



Review Stoichiometry Answer Key Mixed Review:

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In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**Review Stoichiometry Answer Key Mixed Review**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book is central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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