

EARTH AND LIFE SCIENCE
2ND QUARTER
G12 ABM- REVIEWER

LESSON 1:

Exogenic Processes

- It is a **blanket term** for those processes which operate on or close to the surface of the Earth.
- It includes **geological phenomena and processes** that originate externally to the Earth's surface.

Types of Exogenic Processes

Weathering

- It is the **breaking down of rocks into fragments or into smaller size**.
- Every rock had undergone weathering, whether it is **disintegration (physical weathering)** or **decomposition (chemical weathering)**.

Physical Weathering

- Refers to the **breakdown of rocks into pieces without any change in its composition**.

Types of Physical Weathering

Unloading

- It is the **uplifted intrusive igneous rocks**.
- It means rocks from **beneath the surface go above the surface**.
- **Weathering are caused by variety of factors** such as **tectonism and volcanism**.

Thermal Expansion & Contractions

- It refers to the **effect of temperature to rocks**.
- When exposed to **high temperature, rocks expand**; when exposed to **low temperature, rocks contract**.

Freeze-Thaw Weathering

- It is **most common in the North pole and South**.
- It occurs when a **fractured rock absorbs water** through its crack then **got frozen**.
- As mention in Thermal Expansion and contraction, the **colder the temperature, the molecule in it expand**.

Hydration

- It refers to the **amount of water being absorbed by the rock**.
- *When hydrated – too much or plenty of water is present – the rock softens.*
- *When it is dehydrated – too less or lack of water- the rock hardens.*
- The **repeated process of this causes weathering**

Salt Crystal Growth

- It usually **happens in the sea or the seaside**.
- It happens when **salt water comes in contact with rocks**.
- The **wind can also take part in the salt crystal growth**.

Chemical Weathering

- Refers to the **changes in the composition of rock**.

Types of Chemical Weathering

Oxidation

- O^2 in H^2O comes in contact with certain **elements in minerals of rocks**, and **chemical reaction can occur**.
- E.g. Fe and Al - $Fe^2 O^3$ and $Al^2 O^3$

Hydrolysis

- H^2O molecules alone or CO^2 in H^2O react with **chemical composition/components of rock minerals** to create **new components or compounds**.
- E.g. *silicate minerals (containing silica and O^2) in igneous rock silicate minerals – clay minerals*
 $= KaSi^2 O^2 + H^2O - Ha/Si$
- **Reactions that produces weak acid**.

Solution

- **Rock forming mineral dissolve in water**
- E.g. *halite mineral containing rock salt (NaCl) susceptible to solution in H^2O*

Carbonation

- Involves CO^2 and H^2O molecules reacting with, and thereby **decomposing rock material**.
- E.g. *carbonate rocks (CO^2) like limestone (composed of $CaCO^2$) ($O^2 + H^2O - H^2 CO^2$ (carbonic acid) $CaCO$ (limestone) + $H^2 CO^2$ (carbonic acid) + $Ca HCO^2$) calcium bicarbonate*

Mass Wasting

- also called **mass movement**
- **downslope transport of surface materials in direct response to gravity**.
- **heavier objects have greater downward materials in direct response gravity**.
- the **force of gravity encourages rock sediments and soil to move downhill or sloping surfaces**.

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Table of Contents Possible Essays For Life Sciences G12 June Exam

1. Understanding the eBook Possible Essays For Life Sciences G12 June Exam
 - The Rise of Digital Reading Possible Essays For Life Sciences G12 June Exam
 - Advantages of eBooks Over Traditional Books
2. Identifying Possible Essays For Life Sciences G12 June Exam
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Possible Essays For Life Sciences G12 June Exam
 - User-Friendly Interface
4. Exploring eBook Recommendations from Possible Essays For Life Sciences G12 June Exam
 - Personalized Recommendations

- Possible Essays For Life Sciences G12 June Exam User Reviews and Ratings
- Possible Essays For Life Sciences G12 June Exam and Bestseller Lists
- 5. Accessing Possible Essays For Life Sciences G12 June Exam Free and Paid eBooks
 - Possible Essays For Life Sciences G12 June Exam Public Domain eBooks
 - Possible Essays For Life Sciences G12 June Exam eBook Subscription Services
 - Possible Essays For Life Sciences G12 June Exam Budget-Friendly Options
- 6. Navigating Possible Essays For Life Sciences G12 June Exam eBook Formats
 - ePub, PDF, MOBI, and More
 - Possible Essays For Life Sciences G12 June Exam Compatibility with Devices
 - Possible Essays For Life Sciences G12 June Exam Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Possible Essays For Life Sciences G12 June Exam
 - Highlighting and Note-Taking Possible Essays For Life Sciences G12 June Exam
 - Interactive Elements Possible Essays For Life Sciences G12 June Exam
- 8. Staying Engaged with Possible Essays For Life Sciences G12 June Exam
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Possible Essays For Life Sciences G12 June Exam
- 9. Balancing eBooks and Physical Books Possible Essays For Life Sciences G12 June Exam
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Possible Essays For Life Sciences G12 June Exam
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Possible Essays For Life Sciences G12 June Exam
 - Setting Reading Goals Possible Essays For Life Sciences G12 June Exam
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Possible Essays For Life Sciences G12 June Exam
 - Fact-Checking eBook Content of Possible Essays For Life Sciences G12 June Exam

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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