

Polyatomic Ion Worksheet

Place the charges on each ion – Write the Formula – Name the Compound

Remember – The cation always goes first.

| | | | | | | | |
|----------|--------------------|---|-----------------|----------|---|-------|------------------------|
| Example: | NH_4^{+1} | & | I^{-1} | Formula: | <u>NH_4I</u> | Name: | <u>Ammonium Iodide</u> |
| 1. | NH_4 | & | O | Formula: | _____ | Name: | _____ |
| 2. | Ca | & | CO_3 | Formula: | _____ | Name: | _____ |
| 3. | HCO_3 | & | Na | Formula: | _____ | Name: | _____ |
| 4. | NO_3 | & | Mg | Formula: | _____ | Name: | _____ |
| 5. | K | & | SO_4 | Formula: | _____ | Name: | _____ |
| 6. | Li | & | PO_4 | Formula: | _____ | Name: | _____ |

Finding the Formula from the Name

| | | | | | |
|----------|---------------------|-------|--|----------|---------------------------------|
| Example: | Lithium Chloride | Ions: | <u>Li^{+1} & Cl^{-1}</u> | Formula: | <u>LiCl</u> |
| 1. | Sodium Oxide | Ions: | _____ | Formula: | _____ |
| 2. | Potassium Bromide | Ions: | _____ | Formula: | _____ |
| 3. | Calcium Nitride | Ions: | _____ | Formula: | _____ |
| 4. | Beryllium Sulfide | Ions: | _____ | Formula: | _____ |
| 5. | Magnesium Iodide | Ions: | _____ | Formula: | _____ |
| 6. | Aluminum Oxide | Ions: | _____ | Formula: | _____ |
| 7. | Ammonium Bromide | Ions: | _____ | Formula: | _____ |
| 8. | Calcium Bicarbonate | Ions: | _____ | Formula: | _____ |
| 9. | Potassium Phosphate | Ions: | _____ | Formula: | _____ |
| 10. | Aluminum Sulfate | Ions: | _____ | Formula: | _____ |

Polyatomic Ions Worksheet With Answers Physical Science If8767

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Table of Contents Polyatomic Ions Worksheet With Answers Physical Science If8767

1. Understanding the eBook Polyatomic Ions Worksheet With Answers Physical Science If8767
 - The Rise of Digital Reading Polyatomic Ions Worksheet With Answers Physical Science If8767
 - Advantages of eBooks Over Traditional Books
2. Identifying Polyatomic Ions Worksheet With Answers Physical Science If8767
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Polyatomic Ions Worksheet With Answers Physical Science If8767
 - User-Friendly Interface
4. Exploring eBook Recommendations from Polyatomic Ions Worksheet With Answers Physical Science If8767
 - Personalized Recommendations

- Polyatomic Ions Worksheet With Answers Physical Science If8767 User Reviews and Ratings
- Polyatomic Ions Worksheet With Answers Physical Science If8767 and Bestseller Lists
- 5. Accessing Polyatomic Ions Worksheet With Answers Physical Science If8767 Free and Paid eBooks
 - Polyatomic Ions Worksheet With Answers Physical Science If8767 Public Domain eBooks
 - Polyatomic Ions Worksheet With Answers Physical Science If8767 eBook Subscription Services
 - Polyatomic Ions Worksheet With Answers Physical Science If8767 Budget-Friendly Options
- 6. Navigating Polyatomic Ions Worksheet With Answers Physical Science If8767 eBook Formats
 - ePub, PDF, MOBI, and More
 - Polyatomic Ions Worksheet With Answers Physical Science If8767 Compatibility with Devices
 - Polyatomic Ions Worksheet With Answers Physical Science If8767 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Polyatomic Ions Worksheet With Answers Physical Science If8767
 - Highlighting and Note-Taking Polyatomic Ions Worksheet With Answers Physical Science If8767
 - Interactive Elements Polyatomic Ions Worksheet With Answers Physical Science If8767
- 8. Staying Engaged with Polyatomic Ions Worksheet With Answers Physical Science If8767
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Polyatomic Ions Worksheet With Answers Physical Science If8767
- 9. Balancing eBooks and Physical Books Polyatomic Ions Worksheet With Answers Physical Science If8767
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Polyatomic Ions Worksheet With Answers Physical Science If8767
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Polyatomic Ions Worksheet With Answers Physical Science If8767
 - Setting Reading Goals Polyatomic Ions Worksheet With Answers Physical Science If8767
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Polyatomic Ions Worksheet With Answers Physical Science If8767
 - Fact-Checking eBook Content of Polyatomic Ions Worksheet With Answers Physical Science If8767

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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