Periodic Table Extra Practice Worksheet

Directions: Answer the following questions using the Periodic Table below.

IA.							VIII
1 H 1	IIA	IIIB	IVB	VB	VIB	VIIB	2 He 4
3	4	5	6	7	8	9	10
Li	Be	B	C	N	O	F	Ne
7	9	11	12	14	16	19	20
11	12	13	14	15	16	17	18
Na	Mg	Al	Si	P	S	CI	Ar
23	24	27	28	31	32	35	40

1. How many protons, neutrons, and electrons are in each of the following atoms?

Element	Boron (B)	Sodium (Na)	Phosphorus (P)
Protons	5	11.	15
Neutrons	6	12	16
Electrons	5	11	15

2. Which diagram best shows nucleus of beryllium? Circle the correct answer.

$$\begin{pmatrix}
n = 4 \\
e = 5
\end{pmatrix}$$
 $\begin{pmatrix}
p = 4 \\
e = 4
\end{pmatrix}$
 $\begin{pmatrix}
p = 5 \\
n = 4
\end{pmatrix}$
 $\begin{pmatrix}
p = 5 \\
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 $\begin{pmatrix}
p = 4 \\
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\end{pmatrix}$
 $\begin{pmatrix}
p = 9 \\
n = 4
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 $\begin{pmatrix}
p = 4 \\
n = 5
\end{pmatrix}$
 $\begin{pmatrix}
p = 4 \\
n = 4
\end{pmatrix}$
 $\begin{pmatrix}
p = 9 \\
n = 4
\end{pmatrix}$

THE ANWSER IS D

3. Fill in the chart below from the data provided.

ELEMENT	ATOMIC NUMBER	MASS NUMBER	NUMBER OF PROTONS	NUMBER OF NEUTRONS	NUMBER OF ELECTRONS
Aluminum	a.13	27	b.13	c.14	d.13
Hydrogen	e.1	f.1	1	g.0	h.1
Argon	i.18	j.40	k.18	1.22	18
Magnesium	12	m.24	n.12	0.12	p.12

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Daniel F McAuley

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Table of Contents Periodic Table Extra Practice Worksheet Answers

- 1. Understanding the eBook Periodic Table Extra Practice Worksheet Answers
 - The Rise of Digital Reading Periodic Table Extra Practice Worksheet Answers
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Periodic Table Extra Practice Worksheet Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Table Extra Practice Worksheet Answers
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Periodic Table Extra Practice Worksheet Answers
 - Personalized Recommendations
 - Periodic Table Extra Practice Worksheet Answers User Reviews and Ratings
 - Periodic Table Extra Practice Worksheet Answers and Bestseller Lists
- 5. Accessing Periodic Table Extra Practice Worksheet Answers Free and Paid eBooks
 - Periodic Table Extra Practice Worksheet Answers Public Domain eBooks
 - Periodic Table Extra Practice Worksheet Answers eBook Subscription Services
 - Periodic Table Extra Practice Worksheet Answers Budget-Friendly Options
- 6. Navigating Periodic Table Extra Practice Worksheet Answers eBook Formats

- ePub, PDF, MOBI, and More
- Periodic Table Extra Practice Worksheet Answers Compatibility with Devices
- Periodic Table Extra Practice Worksheet Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Table Extra Practice Worksheet Answers
 - Highlighting and Note-Taking Periodic Table Extra Practice Worksheet Answers
 - Interactive Elements Periodic Table Extra Practice Worksheet Answers
- 8. Staying Engaged with Periodic Table Extra Practice Worksheet Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Table Extra Practice Worksheet Answers
- 9. Balancing eBooks and Physical Books Periodic Table Extra Practice Worksheet Answers
 - Benefits of a Digital Library
 - o Creating a Diverse Reading Collection Periodic Table Extra Practice Worksheet Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Table Extra Practice Worksheet Answers
 - Setting Reading Goals Periodic Table Extra Practice Worksheet Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Table Extra Practice Worksheet Answers
 - Fact-Checking eBook Content of Periodic Table Extra Practice Worksheet Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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