

Introduction to the Periodic Table Worksheet

Objective: Use the notes from class, along with your own outside research, to familiarize yourself with the periodic table.

Group	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18																	
1	1 H																	2 He																	
2	3 Li	4 Be											5 B	6 C	7 N	8 O	9 F	10 Ne																	
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4	19 K	20 Ca	21 Sc		22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr																
5	37 Rb	38 Sr	39 Y		40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe																
6	55 Cs	56 Ba	57 La		58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn		
7	87 Fr	88 Ra	89 Ac		90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Rg	112 Cn	113 Nh	114 Fl	115 Mc	116 Lv	117 Ts	118 Og		

1. What is the periodic table and what is it used for?

2. How are elements arranged in the periodic table?

3. What is the difference between a metal and a nonmetal?

4. What is an alkali metal, an alkaline earth metal, a halogen, and a noble gas?

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Mark Steyvers



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