45 grams of nitrogen monoxide (NO) react with an excess of sodium hydroxide (NaOH) following the balanced equation above. 6 grams of water were obtained from the reaction

What is the % yield?

Nitrogen monoxide (NO) gmm:

(1)(14.0) + (1)(16.0) 30.0 g/mo)

Water (H<sub>2</sub>O) gmm:

(2)(1.0) + (1)(16.0) 18.0 g/mol

Yield

45 46 | 18.0 g/mol

Yield

45 46 | 18.0 g/mol

Yield

45 46 | 18.0 g/mol

# **Percent Yield Calculations Answers**

**Stephen DeMeo** 

**Percent Yield Calculations Answers:** 

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