

5 - Periodicity Activity

20 19 18 17 16 15 14 13 12 11 10 9 8 7 6 5 4 3 2 1

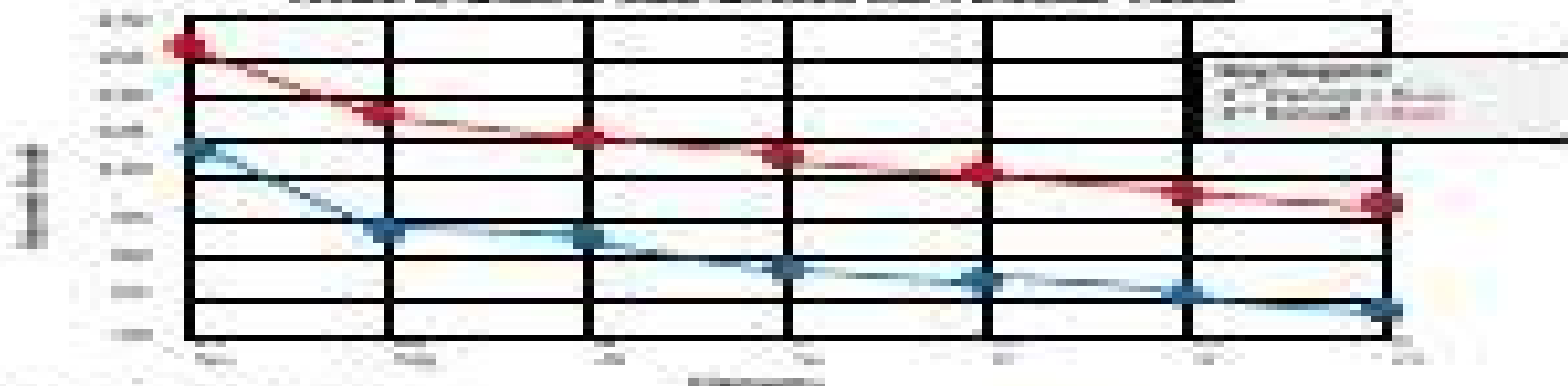
Instructions: Using the graphs above fill the boxes of atomic properties below:

1. Draw a line graph (atomic mass vs) graphing Period 2's data (atomic graph below)
2. Is a slight downward trend in this graph Period 2's data for the atomic graph.
3. Label the line as downward trend
4. Answer all questions...

	1	2	3	4	5	6	7	8
2	Li	Be	B	C	N	O	F	Ne
3	Na	Mg	Al	Si	P	S	Cl	Ar
4	K	Ca	Sc	Ti	V	Cr	Mn	Fe
5	Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru
6	Cs	Ba	La	Hf	Ta	W	Re	Os
7	Fr	Ra	Ac	Rf	Mo	Db	Sg	Bh

Now label the elements above in the following way:

Trend in Atomic Mass Across the Periodic Table



Illustrated Trend Questions

1. Following the trend in atomic mass across the periodic table downward
2. This shows how atomic mass increases across the periodic table. It is a downward trend in atomic mass.

Illustrated Trend Questions

1. This shows how atomic mass increases across the periodic table. It is a downward trend in atomic mass.
2. This shows how atomic mass increases across the periodic table. It is a downward trend in atomic mass.

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Jianjun Gao



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