Periodic Table Named-Groups Worksheet – 2

DIRECTIONS: Use the Periodic Table shown below to answer each question, <u>using the letters in</u> each "box" to identify the required element(s).

A		7							_			
		1							D		E	
C												
	E				G			H				P
	L									I		Г
J		Z		M			N				0	F
o l		\B\		s	т	0	id					
		\.	\	100		20		50 58			110 19	
		1	v				w					
		- 1	×									

List elements in alphabetical order when more than 1 element answers a question.

	ment A is in a box all by itself because: A) it is really a Noble Gas		
gas	 b) it does NOT share the properties of the other elements in its group 	E) it is a liq	uid

 Which of these elements is/are	highly reactive?	A) C	B) "E"	C) . K.	D) A, B, and C	E) A and B only

3) What is the letter of the element whose atomic number is <u>closest</u> to the element labeled "z"?

4) Which letter labels the halogen element with the lowest atomic number?

5) Which of these elements is the least reactive? A) "o" B) "x" C) "A" D) "o" E) "B"

6) Which of these elements is/are <u>NOT</u> transition elements/metals (list ALL that are not transition elements)? A) "6" B) "N" C) "N" D) "N" E) "2"

continued on other side

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J Spring

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- 1. Understanding the eBook Periodic Table Packet Graphs Worksheets Answers
 - The Rise of Digital Reading Periodic Table Packet Graphs Worksheets Answers
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Periodic Table Packet Graphs Worksheets Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Table Packet Graphs Worksheets Answers
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Periodic Table Packet Graphs Worksheets Answers
 - Personalized Recommendations

- Periodic Table Packet Graphs Worksheets Answers User Reviews and Ratings
- Periodic Table Packet Graphs Worksheets Answers and Bestseller Lists
- 5. Accessing Periodic Table Packet Graphs Worksheets Answers Free and Paid eBooks
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 - Periodic Table Packet Graphs Worksheets Answers Enhanced eBook Features
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 - Highlighting and Note-Taking Periodic Table Packet Graphs Worksheets Answers
 - Interactive Elements Periodic Table Packet Graphs Worksheets Answers
- 8. Staying Engaged with Periodic Table Packet Graphs Worksheets Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Table Packet Graphs Worksheets Answers
- 9. Balancing eBooks and Physical Books Periodic Table Packet Graphs Worksheets Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Table Packet Graphs Worksheets Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Table Packet Graphs Worksheets Answers
 - Setting Reading Goals Periodic Table Packet Graphs Worksheets Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Table Packet Graphs Worksheets Answers
 - Fact-Checking eBook Content of Periodic Table Packet Graphs Worksheets Answers

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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