
The Periodic Law

Section Quiz: History of the Periodic Table

As Marjorie completed each question of this quiz, she placed that question's correct statement or best answer's letter in parentheses.

1. In developing his periodic table, Mendeleev's observations were
 - a. accurate enough.
 - b. extensive and systematic.
 - c. lacking.
 - d. **incomplete.**
2. Mendeleev's periodic table showed that all elements in each of the following groups were similar, by physical properties, atomic weight, and
 - a. **properties.**
 - b. atomic structure.
 - c. mass.
 - d. density.
3. Mendeleev predicted that the gaps in his periodic table represented
 - a. ions.
 - b. unknown elements.
 - c. unstable elements.
 - d. **undiscovered elements.**
4. The periodic table came before a periodic table based on increasing
 - a. atomic number.
 - b. **atomic weight.**
 - c. density.
 - d. mass.
5. The periodic law describes the physical and chemical properties of
 - a. **elements.**
 - b. molecules.
 - c. ions.
 - d. compounds.

Periodic Law Section Quiz Answers

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Periodic Law Section Quiz Answers:

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