

Chemistry: Periodic Table Unit Study Guide

Objectives:

1. Define the following terms:
 - Period
 - Group
 - Valence
 - Atomic Number
 - Mass Number
 - Isotope
2. Draw the Periodic Table.
3. Label the following families on the periodic table:
 - Alkali Metals
 - Alkaline Earth Metals
 - Transition Metals
 - Metals
 - Nonmetals
 - Halogens
 - Noble Gases
4. An element has 11 protons and 11 electrons. What is its atomic number?
5. An element has 11 protons and 10 electrons. What is its charge?
6. An element has 11 protons and 12 neutrons. What is its mass number?
7. An element has 11 protons and 11 neutrons. What is its mass number?
8. An element has 11 protons and 11 neutrons. What is its mass number?
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100. An element has 11 protons and 11 neutrons. What is its mass number?

Name: _____ Date: _____ Period: _____

Periodic Table Unit Study Guide

Objectives and Overview of the Periodic Table

1. What are the main groups of elements in the periodic table?

- Alkali Metals
- Alkaline Earth Metals
- Transition Metals
- Metals
- Nonmetals
- Halogens
- Noble Gases

2. Label the following families on the periodic table: Alkali Metals, Alkaline Earth Metals, Transition Metals, Metals, Nonmetals, Halogens, and Noble Gases. For each family, write down the color designation and the number of groups it contains.

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
H	He																
Li	Be	B	C	N	O	F	Ne										
Na	Mg	Al	Si	P	S	Cl	Ar										
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
Cs	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn
Fr	Ra																

Alkali Metals

Alkaline Earth Metals

Transition Metals

Halogens

Noble Gases

3. Match the periodic table color number of groups with the color designation family.

4. Fill in the following chart:

Element	Astatine	Ca^{2+}	$1s^2 2s^2 2p^6 3s^2$	
Number of Electrons				
Group Number (if applicable)				
Period Number (if applicable)				
Atomic Number (usually 1-118)				

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JR Anderson



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