# **Periodic Table Extra Practice Worksheet**

Directions: Answer the following questions using the Periodic Table below.

| IA.            |                |                |                |               |               |                | VIII           |
|----------------|----------------|----------------|----------------|---------------|---------------|----------------|----------------|
| 1<br>H<br>1    | IIA            | IIIB           | IVB            | VB            | VIB           | VIIB           | 2<br>He<br>4   |
| 3<br>Li<br>7   | Be<br>9        | 5<br>B<br>11   | 6<br>C<br>12   | 7<br>N<br>14  | 8<br>O<br>16  | 9<br>F<br>19   | 10<br>Ne<br>20 |
| 11<br>Na<br>23 | 12<br>Mg<br>24 | 13<br>Al<br>27 | 14<br>Si<br>28 | 15<br>P<br>31 | 16<br>S<br>32 | 17<br>CI<br>35 | 18<br>Ar<br>40 |

1. How many protons, neutrons, and electrons are in each of the following atoms?

| Element   | Boron (B) | Sodium (Na) | Phosphorus (P) |
|-----------|-----------|-------------|----------------|
| Protons   | 5         | 11.         | 15             |
| Neutrons  | 6         | 12          | 16             |
| Electrons | 5         | 11          | 15             |

2. Which diagram best shows nucleus of beryllium? Circle the correct answer.

$$\begin{pmatrix}
n = 4 \\
e = 5
\end{pmatrix}$$
 $\begin{pmatrix}
p = 4 \\
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\end{pmatrix}$ 
 $\begin{pmatrix}
p = 5 \\
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\end{pmatrix}$ 
 $\begin{pmatrix}
p = 5 \\
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\end{pmatrix}$ 
 $\begin{pmatrix}
p = 4 \\
n = 5
\end{pmatrix}$ 
 $\begin{pmatrix}
p = 9 \\
n = 4
\end{pmatrix}$ 
 $\begin{pmatrix}
p = 4 \\
n = 5
\end{pmatrix}$ 
 $\begin{pmatrix}
p = 4 \\
n = 4
\end{pmatrix}$ 
 $\begin{pmatrix}
p = 9 \\
n = 4
\end{pmatrix}$ 

#### THE ANWSER IS D

3. Fill in the chart below from the data provided.

| ELEMENT   | ATOMIC<br>NUMBER | MASS<br>NUMBER | NUMBER OF<br>PROTONS | NUMBER OF<br>NEUTRONS | NUMBER OF<br>ELECTRONS |
|-----------|------------------|----------------|----------------------|-----------------------|------------------------|
| Aluminum  | a.13             | 27             | b.13                 | c.14                  | d.13                   |
| Hydrogen  | e.1              | f.1            | 1                    | g.0                   | h.1                    |
| Argon     | i.18             | j.40           | k.18                 | 1.22                  | 18                     |
| Magnesium | 12               | m.24           | n.12                 | 0.12                  | p.12                   |

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**VM** Jensen

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