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Exceed to Succeed

Worksheet: Periodic Trends

1. ATOMIC RADIUS

For each of the following sets of atoms, rank the atoms from smallest to largest atomic radius.

- a. Li, C, F
- b. Li. Na. K
- c. Ge. P. O.
- d. C. N. Al
- e. Al. Cl. Ga.

IONIC RADIUS 2...

For each of the following sets of ions, rank them from smallest to largest ionic radius.

- a. Mg²⁺, Si⁴⁻, S²⁻ b. Mg²⁺, Ca²⁺, Ba²⁺
- c. F. Cl. Br
- d. Ba²⁺, Cu²⁺, Zn²⁺ e. Si⁴⁻, P³⁻, O²⁻

IONIZATION ENERGY 3...

For each of the following sets of atoms, rank them from lowest to highest ionization energy.

- a. Mg, Si, S
- b. Mg, Ca, Ba
- c. F. Cl. Br
- d. Ba. Cu. Ne
- e. Si. P. He

44. ELECTRONEGATIVITY

For each of the following sets of atoms, rank them from lowest to highest electronegativity.

- a. Li, C, N
- b. C. O. Ne
- c. Si. P. O
- d. K. Mg, P
- e. S. F. He

Periodic Trends Extension Questions Worksheet

Mike Jess

Periodic Trends Extension Questions Worksheet:

Unveiling the Magic of Words: A Review of "Periodic Trends Extension Questions Worksheet"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is actually aweinspiring. Enter the realm of "**Periodic Trends Extension Questions Worksheet**," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound affect the souls of its readers.

https://crm.avenza.com/data/browse/default.aspx/physical%20science%20final%20review%20answers.pdf

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