

# CECIL CONVENT SCHOOL

## PERIODIC ASSESSMENT-2

Class 11 - Chemistry

Time Allowed: 1 hour

Maximum Marks: 20

General Instructions:

1. There are 11 questions in this question paper.
2. SECTION A - Q. No. 1 to 5 are very short answer questions carrying 1 mark each.
3. SECTION B - Q. No. 6 to 8 are short answer questions carrying 2 marks each.
4. SECTION C- Q. No. 9 to 11 is case based question carrying 3 marks each.
5. All questions are compulsory.
6. Attempt questions in order.

1. Water can be lifted into the water tank at the top of the house with the help of a pump. Then why is it not considered to be spontaneous? [1]
2. Neither  $q$  nor  $W$  is a state function but  $q + W$  is a state function. Explain why? [1]
3. What is the enthalpy change for an adiabatic process? [1]
4. State the law of chemical equilibrium. [1]
5. A tank is full of water. Water is coming in as well as going out at same rate. What will happen to level of water in a tank? What is name given to such state? [1]
6. Calculate the standard enthalpy of formation of  $\text{CH}_3\text{OH}(\text{l})$  from the following data: [2]



7. The equilibrium constant for a reaction is 10. What will be the value of  $\Delta G^\ominus$ ?  $R = 8.314 \text{ JK}^{-1} \text{ mol}^{-1}$ ,  $T = 300 \text{ K}$ . [2]
8. Write the equilibrium constant ( $K_c$ ) expression for the following reactions. [2]



9. i. Why standard entropy of an elementary substance is not zero whereas standard enthalpy of formation is taken as zero? [3]  
 ii. Under what conditions will the reaction occur, if  
 a. both  $\Delta H$  and  $\Delta S$  are positive  
 b. both  $\Delta H$  and  $\Delta S$  are negative
10. Dihydrogen gas is obtained from the natural gas by partial oxidation with steam as per following endothermic reaction: [1]  
 $\text{CH}_4(\text{g}) + \text{H}_2\text{O}(\text{g}) \rightleftharpoons \text{CO}(\text{g}) + 3\text{H}_2(\text{g})$   
 a. Write the expression for  $K_p$  for the above reaction.

# Periodic Assessment 2 Chemistry Answers

**JE Gale**



## **Periodic Assessment 2 Chemistry Answers:**

## Reviewing **Periodic Assessment 2 Chemistry Answers**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Periodic Assessment 2 Chemistry Answers**," an enthralling opus penned by a highly acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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