Mole Calculation Practice Worksheet

Answer the following questions:



1) How many moles are in 25 grams of water?

> 1.4 moles 250 H20 1 mai H20 -

2) How many grams are in 4.5 moles of Li₂O?

130 0

4.5 ms/Li₂0 | 39.88₃Li₂0 _ | Imil U.O | How many molecules are in 23 moles of oxygen?

31

1.4 × 1025 moleculos

23 Mol O2 4.02 × 103 motocules =

How many moles are in 3.4 x 1023 molecules of H-SO4? 4)

0.56 moles

3.4×10°3 molecules H. Soy | 1 mol H. Soy =

5) How many molecules are in 25 grams of NH₃?

8.8 ×1023 molecules

259 NH3 | ImeINHs | 6.02 x10 3 MI Becules NH3 =

How many grams are in 8.2 x 10²² molecules of N₂I₂? 6)

8.2 × 102 2 molecules No Iv | 1 mol N. Iv | 1789,423 No Iv =

Practice Equations In Moles

Thomas Bond, Chris Hughes

Practice Equations In Moles:

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