

Name _____

Physical Science Chemical Bonding Study Guide

A. Covalent & Ionic Compounds/Bonds

1. Which ion always forms with a charge of 1-? _____ (halogens)
which cation is always with a charge of 1+? _____ (alkali)
2. How electrons in the halogen group has _____ valence electrons, and needs to _____ 1 more to have 8. This is because of the _____ rule.
3. How _____ bond in the alkaline elements oppositely charged ions.
How is ionic bond is _____
4. On a covalent bond, atoms _____ electrons, whereas, ionic bond is an ionic bond, atoms _____ electrons.
5. Elements group 16 has 6 valence electrons. It is usually found as _____, and when they're bonded usually to be _____
How can group 16 be distinguished in terms, color, shape, and molecular formula by
looking at the _____
6. How atoms can have up to _____ valence electrons. This usually happens in the periodic table that all have 8 valence electrons is called _____
7. Atoms can gain, lose, or share _____ by forming ionic bonds.

B. Formulas for Ionic Compounds - Fill in the blank below

Write the correct chemical formula	Aluminum	Fluorine	Phosphorus	Chlorine
COVALENT BONDING IS				

1. Write the chemical formula for each of the following ions:
Cation: _____

Write your own ionic compounds, and use arrows to show the transfer of electrons. Use your periodic table to find their oxidation numbers. They will combine so that their oxidation numbers cancel out to zero. Use the criss cross method if their oxidation numbers do not cancel.

Physical Science Chemical Bonding Study Guide

Ensheng Dong



Physical Science Chemical Bonding Study Guide:

Reviewing **Physical Science Chemical Bonding Study Guide**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Physical Science Chemical Bonding Study Guide**," an enthralling opus penned by a very acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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