

T.Y.B.Sc. Semester 5 Mock Exam Question Paper
Chemistry Paper I – Physical Chemistry
Multiple Choice Questions

- Q1 Vapour pressure of a liquid solventwhen a non volatile solute is dissolve in it. 1M
a) Increase
b) Decrease
c) no change
d) Expand
- Q2 Raoult's law is expressed by.....
a) $\frac{P^o - P}{P^o} = X_2$
b) $\frac{P - P^o}{P^o} = X_2$
c) $\frac{P - P^o}{P} = X_2$
d) $\frac{P - P^o}{P^o} = X$
- Q3 is used in Rast method because it has a high value of K_f .
a) Camphor
b) Bi phenyl
c) Naphthalene
d) Sodium chloride
- Q4 Boiling point elevation of the solution ΔT_b is given as.....
a) $\Delta T_b = T - T_o$
b) $\Delta T_b = T_o - T$
c) $\Delta T_b = T + T_o$
d) $\Delta T_b = T \times T_o$
- Q5 The unit for activation of energy is
a) kJ mol^{-1}
b) kJ K^{-1}
c) $\text{kJ K}^{-1} \text{mol}^{-1}$
d) kJ R^{-1}

Physical Chemistry Multiple Choice Questions

Singhal



Physical Chemistry Multiple Choice Questions:

Multiple Choice Questions in Physical Chemistry for Undergraduate Students Dr Kusum Baser, 2022-08-25

Objective questions are very much useful for practice of particular topic Multiple choice questions are important to practice any topic to solve objective questions very deep study and knowledge of specific topic is needed This book helps the students to do practice in such a way so they can do preparation for competitive exams This book covers important topics of physical chemistry for undergraduate students *Physical Chemistry: A Level Multiple Choice Questions* Margaret Marion Langley Oblitas, 1970-01-01

Physical Chemistry M.M.L. Oblitas, 1971 *Physical Chemistry* M.M.L. Oblitas, 1977 **The Pearson Guide to Physical Chemistry for the IIT JEE** Singhal, Sharma K.K. & Sharma L.K., 2016 **A Textbook of Physical Chemistry** **Atkins' Physical Chemistry** Peter William Atkins, Julio De Paula, James Keeler, 2023 Revised edition of Atkins Physical chemistry Peter Atkins Julio de Paula James Keeler Eleventh edition 2018

Elements of Physical Chemistry Peter Atkins, Julio de Paula, 2013 *Elements of Physical Chemistry* has been carefully crafted to help students increase their confidence when using physics and mathematics to answer fundamental questions about the structure of molecules how chemical reactions take place and why materials behave the way they do Class 9 Chemistry MCQ (Multiple Choice Questions) Arshad Iqbal, The Class 9 Chemistry Multiple

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AQA A-level Year 2 Chemistry Student Guide: Physical chemistry 2 Alyn G. McFarland, Nora Henry, 2016-04-04 Exam Board AQA Level A level Subject Chemistry First Teaching September 2016 First Exam June 2017 Written by experienced examiners Alyn McFarland and Nora Henry this Student Guide for Chemistry Helps students identify what they need to know with a concise summary of the topics examined in the AS and A level specifications Consolidates understanding with tips and knowledge check questions Provides opportunities to improve exam technique with sample answers to exam style questions

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Pharmaceutical Physical Chemistry: Theory and Practices S. K. Bhasin, Physical chemistry is a compulsory paper offered to all the students of pharmacy There is a dearth of good books that exclusively cover the syllabi of physical chemistry offered to pharmacy courses Pharmaceutical Physical Chemistry Theory and Practices has been designed considering their requirements laid down by AICTE and other premier institutes universities Apart from the theory 20 most common laboratory experiments have been included to make this book a unique offering to the students of pharmacy **AQA AS/A Level Year**

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Questions in A-level Chemistry B. P. Bourne, 1969 Essentials of Physical Chemistry 28th Edition Bahl Arun/ Bahl B.S. & Tuli G.D., 2022-03 Essentials of Physical Chemistry is a classic textbook on the subject explaining fundamentals concepts with discussions illustrations and exercises With clear explanation systematic presentation and scientific accuracy the book not only helps the students clear misconceptions about the basic concepts but also enhances students ability to analyse and systematically solve problems This bestseller is primarily designed for B Sc students and would equally be useful for the aspirants of medical and engineering entrance examinations **Elements of Physical Chemistry** Peter William

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Seiffert, 2023-01-30 This book introduces the concepts of physical chemistry of polymers in a format targeted for a blended learning approach It provides a basis to bridge polymer chemistry which targets microscopic chain structures and polymer engineering which targets macroscopic material properties and functions Topics covered are single chain statistics multi chain interactions and chain dynamics both from a viewpoint of structure properties mostly mechanical ones and their interrelation In all that the author encourages the reader to think conceptually Explains complex facts through simplifying models diagrams and illustrations Accessible to chemists chemical engineers materials scientists and physicists Tailored content for an interactive blended learning format *Cracking the GRE Chemistry Test*, 2002 The Princeton Review realizes

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