

7. Give a recipe for preparing a sodium carbonate/sodium bicarbonate buffer solution having a pH of 9.5. **For carbonic acid, H_2CO_3 , $\text{pK}_{\text{a}1} = 6.35$ and $\text{pK}_{\text{a}2} = 10.33$**

8. A saturated solution of silver chromate (Ag_2CrO_4 , a sparingly soluble solid, formula mass 331.0 g/mol), was found to have a chromate ion concentration of $3.2 \times 10^{-4} \text{ M}$ at 25°C . What is (a) the K_{sp} ; and (b) solubility of the salt in mg/mL? **How could you dissolve a sample of the salt?**

9. Calculate the solubility in mg/mL of PbCl_2 (formula mass 276.2 g/mol) in 0.25 M sodium chloride solution at 25°C . K_{sp} for lead chloride is 1.8×10^{-10} at 25°C .

10. Write an equation to show the formation of the complex ion $[\text{Ag}(\text{NH}_3)_2]^+$ from silver ion and aqueous ammonia then calculate the concentration of silver ions in solution if exactly 0.100 mol of silver nitrate is added to exactly **500 mL of 0.800 M ammonia** solution. K_f for $[\text{Ag}(\text{NH}_3)_2]^+$ is 1.7×10^7 .

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M Walker



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Whispering the Techniques of Language: An Psychological Quest through **Recipe For Sodium Carbonate Sodium Bicarbonate Buffer**

In a digitally-driven world where displays reign supreme and instant conversation drowns out the subtleties of language, the profound secrets and emotional nuances hidden within phrases usually get unheard. Yet, located within the pages of **Recipe For Sodium Carbonate Sodium Bicarbonate Buffer** a charming literary prize blinking with organic feelings, lies an extraordinary journey waiting to be undertaken. Composed by an experienced wordsmith, this enchanting opus encourages visitors on an introspective trip, gently unraveling the veiled truths and profound influence resonating within the material of every word. Within the emotional depths with this touching evaluation, we can embark upon a heartfelt exploration of the book is key themes, dissect its interesting publishing design, and yield to the powerful resonance it evokes deep within the recesses of readers hearts.

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