

AP Chemistry Chapter 20 Electrochemistry Practice Free Response #1



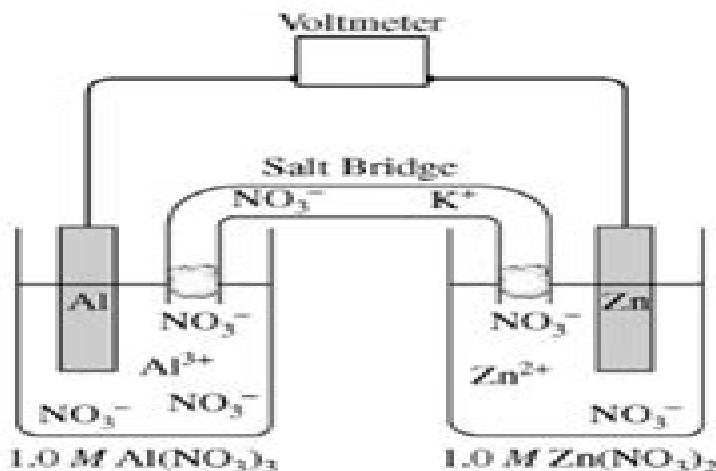
Respond to the following statements and questions that relate to the species and the reaction represented above.

(a) Write the complete electron configuration (e.g.,  $1s^2 2s^2 \dots$ ) for  $\text{Zn}^{2+}$ .

(b) Which species,  $\text{Zn}$  or  $\text{Zn}^{2+}$ , has the greater ionization energy? Justify your answer.

(c) Identify the species that is oxidized in the reaction.

The diagram shows a galvanic cell based on the reaction. Assume that the temperature is  $25^\circ\text{C}$ .



(d) The diagram includes a salt bridge that is filled with a saturated solution of  $\text{KNO}_3$ . Describe what happens in the salt bridge as the cell operates.

(e) Determine the value of the standard voltage,  $E^\circ$ , for the cell.

# Practice Hall Chemistry Chapter 20

**Annelies Wilder-Smith**



## **Practice Hall Chemistry Chapter 20:**

## Enjoying the Track of Phrase: An Psychological Symphony within **Practice Hall Chemistry Chapter 20**

In a world taken by screens and the ceaseless chatter of fast transmission, the melodic elegance and mental symphony created by the written word often fade into the back ground, eclipsed by the constant noise and distractions that permeate our lives. But, set within the pages of **Practice Hall Chemistry Chapter 20** a wonderful literary value overflowing with natural thoughts, lies an immersive symphony waiting to be embraced. Constructed by an outstanding musician of language, that interesting masterpiece conducts visitors on an emotional journey, well unraveling the concealed songs and profound affect resonating within each cautiously crafted phrase. Within the depths of this poignant review, we will investigate the book is key harmonies, analyze its enthralling writing type, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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