

Physical properties of solutions

Even though “like dissolves like”, the presence of **solute molecules in the solvent**

disrupts the bulk solvent intermolecular forces to some extent.

This change in forces means that the **physical properties of solutions**,

like freezing and boiling points, **are slightly different than those of the pure solvent**.

Physical Properties Of Solutions Concept Review

Jessica J Manson



Physical Properties Of Solutions Concept Review:

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