

# Memorandum Grade 11 p/sciences

Physical Sciences Gr11/P2

2  
NSC

September

## QUESTION 1

1.1 A ✓✓

(2)

1.2 B ✓✓

(2)

1.3 C & ✓✓

(2)

1.4 B & ✓✓

(2)

1.5 C & ✓✓

(2)

1.6 D ✓✓

(2) ~~(10)~~ (12)

## QUESTION 2

2.1 The mass of one mole of a substance measured in  $\text{g}\cdot\text{mol}^{-1}$  ✓✓ (2 or 0)

(2)

2.2

Marking criteria		
(a) Any formula $m = nM$ or $N = nN_A$ ✓		
(b) Substitution of $3,114 \times 10^{24}$ and $6,02 \times 10^{23}$ into $N = nN_A$ ✓		
(c) Substitution of $1 \text{ g}\cdot\text{mol}^{-1}$ into $m = nM$ ✓		
(d) Mole ratio 3:6:1 ✓		
(e) $X = 3$ ✓, $Y = 6$ ✓, $Z = 1$ ✓		
C	H	O
$n = \frac{N}{N_A}$ ✓ $n = \frac{3,114 \times 10^{24}}{6,02 \times 10^{23}}$ ✓ $n = 5,173 \text{ mol}$	$n = \frac{m}{M}$ $n = \frac{10,345}{1}$ ✓ $n = 10,345 \text{ mol}$	$n = 1,724 \text{ mol}$
$\frac{5,173}{1,724} = 3$	$\frac{10,345}{1,724} = 6$	$\frac{1,724}{1,724} = 1$
3:6:1 ✓		
$X = 3$ ✓	$Y = 6$ ✓	$O = 1$ ✓

Accept values indicated in the empirical formula:  $\text{C}_3\text{H}_6\text{O}$  ✓✓✓

(7)

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## **Table of Contents Physical Science Memoradam Grade 11 September 2013**

1. Understanding the eBook Physical Science Memoradam Grade 11 September 2013
  - The Rise of Digital Reading Physical Science Memoradam Grade 11 September 2013
  - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Memoradam Grade 11 September 2013
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Physical Science Memoradam Grade 11 September 2013
  - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Memoradam Grade 11 September 2013
  - Personalized Recommendations
  - Physical Science Memoradam Grade 11 September 2013 User Reviews and Ratings
  - Physical Science Memoradam Grade 11 September 2013 and Bestseller Lists
5. Accessing Physical Science Memoradam Grade 11 September 2013 Free and Paid eBooks

- Physical Science Memoradam Grade 11 September 2013 Public Domain eBooks
  - Physical Science Memoradam Grade 11 September 2013 eBook Subscription Services
  - Physical Science Memoradam Grade 11 September 2013 Budget-Friendly Options
6. Navigating Physical Science Memoradam Grade 11 September 2013 eBook Formats
- ePub, PDF, MOBI, and More
  - Physical Science Memoradam Grade 11 September 2013 Compatibility with Devices
  - Physical Science Memoradam Grade 11 September 2013 Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Physical Science Memoradam Grade 11 September 2013
  - Highlighting and Note-Taking Physical Science Memoradam Grade 11 September 2013
  - Interactive Elements Physical Science Memoradam Grade 11 September 2013
8. Staying Engaged with Physical Science Memoradam Grade 11 September 2013
- Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Physical Science Memoradam Grade 11 September 2013
9. Balancing eBooks and Physical Books Physical Science Memoradam Grade 11 September 2013
- Benefits of a Digital Library
  - Creating a Diverse Reading Collection Physical Science Memoradam Grade 11 September 2013
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Physical Science Memoradam Grade 11 September 2013
- Setting Reading Goals Physical Science Memoradam Grade 11 September 2013
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Science Memoradam Grade 11 September 2013
- Fact-Checking eBook Content of Physical Science Memoradam Grade 11 September 2013
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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