

MULTIPLE CHOICE QUESTIONS

Oxidation Number

- Oxidation is a process of
- loss of electrons
 - gain of electrons
 - increase in the negative valency
 - decrease in positive valency.
- When zinc is added to CuSO_4 solution, copper is precipitated, it is so because of
- Reduction of Zn
 - Reduction of Cu^{2+}
 - Hydrolysis of CuSO_4
 - Reduction of SO_4^{2-} .
- The process in which oxidation number increases is
- Reduction
 - Hydrolysis
 - Oxidation
 - Decomposition.
- When zinc metal is added to dilute H_2SO_4 solution, hydrogen is evolved. In this zinc undergoes
- Reduction
 - Oxidation
 - Dissolution
 - None of the above.
- In which of the following, oxidation number of chlorine is +5 ?
- Cl_2O_7
 - ClO_3^-
 - ClO^-
 - ClO_4^- .
- Carbon has zero oxidation number in
- CH_4
 - CH_3Cl
 - CCl_4
 - CH_2Cl_2 .
- The oxidation number of carbon in $\text{C}_{12}\text{H}_{22}\text{O}_{11}$ is
- 0
 - +22
 - +6
 - 6.
- In the following
- $$3\text{Br}_2 + 6\text{CO}_3^{2-} + 3\text{H}_2\text{O} \longrightarrow 5\text{Br}^- + \text{BrO}_3^- + 6\text{HCO}_3^-$$
- Bromine is oxidised and carbonate is reduced
 - Bromine is reduced and carbonate is oxidised

- Bromine is neither reduced nor oxidised
 - Bromine is reduced and oxidised.
9. The oxidation number of C in CH_4 , CH_3Cl , CH_2Cl_2 , CHCl_3 and CCl_4 is respectively
- 4, -2, 0, +2, +4
 - +2, 4, 0, -2, -4
 - 4, 2, 0, -2, -4
 - 0, 2, -2, 4, 4.
10. Chlorine is in +3 oxidation state in
- HCl
 - HClO_4
 - ICl
 - ClF_3 .
11. Oxidation number of Mn in K_2MnO_4 is
- 2
 - 4
 - 6
 - 7.
12. A student states that burning of lime in air is an oxidation process. The reason he gives is that an oxide of the metal is produced on burning. Which one is correct ?
- The statement and reason are true, the reason is correct explanation
 - The statement and reason are true, the explanation is not correct
 - The statement is true but the reason is false
 - The statement is false but the reason is true.
13. For the following reaction in the acidic solution
- $$\text{MnO}_4^- + 8\text{H}^+ + 5e^- \longrightarrow \text{Mn}^{2+} + 4\text{H}_2\text{O},$$
- which of the following gives the true oxidation numbers of the manganese on each side of the equation ?
- +7 to +6
 - +7 to +2
 - +4 to +2
 - 1 to +2.
14. In which of the following compounds, iron has the lowest oxidation number ?
- $\text{Fe}(\text{CO})_5$
 - Fe_2O_3
 - $\text{K}_4[\text{Fe}(\text{CN})_6]$
 - $\text{FeSO}_4(\text{NH}_4)_2\text{SO}_4 \cdot 6\text{H}_2\text{O}$.
15. The oxidation number and covalency of sulphur in sulphur molecule (S_8) are

Answers

- | | | | | | | | | | |
|---------|---------|---------|---------|--------|--------|--------|--------|--------|---------|
| 1. (A) | 2. (B) | 3. (C) | 4. (B) | 5. (B) | 6. (D) | 7. (A) | 8. (D) | 9. (A) | 10. (D) |
| 11. (C) | 12. (D) | 13. (B) | 14. (A) | | | | | | |

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