

1. Write the group numbers in the boxes along the top of the periodic table.
2. Write the period numbers in the boxes down the side of the periodic table.

[illegible]

Elements in the same group have similar properties.

Lithium (Li) and potassium (K) are metals. They are solids at room temperature, have low melting points compared to other metals and are very reactive.

3. Name two other elements that have similar properties to lithium and potassium.

Any two from:

- sodium (Na)
- rubidium (Rb)
- caesium (Cs)
- francium (Fr)

Neon (Ne) and xenon (Xe) are non-metals. They are gases at room temperature and are unreactive.

4. Name two other elements that have similar properties to neon and xenon.

Any two from:

- helium (He)
- argon (Ar)
- krypton (Kr)
- radon (Rn)

Reinforcement Periodic Table Worksheet Answers

Chapter 1

Y Pai



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