

Regents Chemistry
Unit 8 Thermochemistry Practice Test

- At which temperature do both the solid and the liquid phase of water exist?
(1) 273°C (3) 0°C
(2) 100°C (4) 373°C
- What is the freezing point of sodium?
(1) 371°C (3) -98°C
(2) 644°C (4) 98°C
- A 15.75-g piece of iron absorbs 1086.75 joules of heat energy, and its temperature changes from 25°C to 175°C . What is the specific heat capacity of iron?
(1) $1.0\text{ J/g}^{\circ}\text{C}$ (3) $0.46\text{ J/g}^{\circ}\text{C}$
(2) $4.18\text{ J/g}^{\circ}\text{C}$ (4) $2.76\text{ J/g}^{\circ}\text{C}$
- Which statement concerning propanone and water at 50°C is true?
(1) Propanone has a lower vapor pressure and weaker intermolecular forces than water.
(2) Propanone has a lower vapor pressure and stronger intermolecular forces than water.
(3) Propanone has a higher vapor pressure and weaker intermolecular forces than water.
(4) Propanone has a higher vapor pressure and stronger intermolecular forces than water.
- Which phase change is an exothermic process?
(1) $\text{CO}_2(\text{s}) \rightarrow \text{CO}_2(\text{g})$ (2) $\text{Cu}(\text{s}) \rightarrow \text{Cu}(\text{l})$
(3) $\text{NH}_3(\text{l}) \rightarrow \text{NH}_3(\text{g})$ (4) $\text{Hg}(\text{l}) \rightarrow \text{Hg}(\text{s})$
- What is the vapor pressure of water at 105°C ?
(1) 100 kPa (3) 101.3 kPa
(2) 110 kPa (4) 120 kPa
- As a substance changes from a liquid to a gas, the average distance between molecules
(1) decreases
(2) increases
(3) remains the same

- The graph below represents the uniform heating of a substance, starting with the substance as a solid below its melting point.



Which line segment represents an increase in average kinetic energy and *no* change in potential energy?

- (1) BC (3) ED
(2) AB (4) DE
- How many joules are required to melt 100. grams of water?
(1) 226,000 kJ (2) 22,600 J
(3) 33,400 kJ (4) 33,400 J
- What occurs when a 35-gram aluminum cube at $100.^{\circ}\text{C}$ is placed in 90. grams of water at 25°C in an insulated cup?
(1) Heat is transferred from the aluminum to the water, and the temperature of the water decreases.
(2) Heat is transferred from the aluminum to the water, and the temperature of the water increases.
(3) Heat is transferred from the water to the aluminum, and the temperature of the water decreases.
(4) Heat is transferred from the water to the aluminum, and the temperature of the water increases.
- What is the boiling point of propanone if the pressure on its surface is 48 kilopascals?
(1) 25°C (3) 35°C
(2) $30.^{\circ}\text{C}$ (4) $40.^{\circ}\text{C}$

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B Lingard



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