Reference Tables Scavenger Hunt Answers

Name

Directions: Using the Reference Tables for Chemistry, locate the following information.

- Name C<sub>5</sub>H<sub>12</sub> Pentane Table P + R
- Write the equation for the decay of Kr-85. <sup>85</sup>Kr → β + <sup>85</sup>Rb Table N
- Explain how you know that NaPO<sub>4</sub> is soluble in water but NiCrO<sub>4</sub> is not. Na group 1/ all chromates insoluble Table F
- What is the definition of STP, and give the values? Standard Temperature (0oC or 273K) and Pressure (1 atm or 101.3 kPa) Table A
- 5. Name, and give the formulas of the strongest and weakest bases. NaOH / NHs Table L.
- Name C<sub>2</sub>H<sub>3</sub>O<sup>2</sup> or CH<sub>3</sub>COO<sup>3</sup> acetate ion Table E
- What is the solubility of sulfur dioxide at 40 degrees Celsius, in grams? -5g Table G
- What is the freezing point of fluorine? 53 K Table S
- What are the units for the heat of fusion, and what do they mean? J/g heat lost per gram of substance Table B
- What is the symbol for the mole? mol Table D.
- What is the vapor pressure of water at 75°C? 38 kPa Table H.
- How much heat does it take to convert 20g of water to steam at 100°C? 45200 J (45.2 kJ)
- 13. What is the prefix for 1/1000 of a meter? Milli- Table C
- 14. What is the molecular formula of ammonia? NHo Table L.
- What is the formula for the permanganate ion? MnO<sub>4</sub>- Table E.
- 16. Name CH3COOH ethanoic acid (acetic acid) Table R
- Write the symbol for a positron? β+ Table O
- 18. What is the half-life of Pu-239? 2.4 x 104 years Table N
- Is the formation of water, from its elements endothermic or exothermic? exothermic Table I
- 20. What is the atomic mass of silver? 108 g/mol P-Table
- 21. How much heat is released when LiBr dissolves in water? -48.83 kJ Table I
- 22. Give the names and formulas of the strongest and weakest acids. HCl / CH3COOH Table K
- 23. What is the general formula for alkynes? What does it mean? College Table Q.

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**Yijin Wang** 

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