Unit 5, Activity 5

Balancing Redox Reactions using half-reaction method worksheet

Balance the following equations using the half-reaction method. Identify the reactant being oxidized and reduced.

a.
$$Cu^{2+} + I^{1-} \rightarrow Cu + I_3^{1-}$$
 (neutral)

b.
$$MnO_4^{1-} + Ag \rightarrow Mn^{2+} + Ag^{1+}$$
 (acidic)

c.
$$Hg + NO_3^{1-} + Cl^{1-} \rightarrow HgCl_4^{2-} + NO_2$$
 (acidic)

d.
$$AsH_3 + Zn^{2+} \rightarrow H_3AsO_4 + Zn$$
 (acidic)

e.
$$CN^{1-} + MnO_4^{1-} \rightarrow CNO^{1-} + MnO_2$$
 (basic)

g.
$$ClO^{1-} + CrO_2^{1-} \rightarrow CrO_4^{2-} + Cl_2$$
 (basic)

Redox Reaction Exercise And Answers

Thomas Griffiths

Redox Reaction Exercise And Answers:

Whispering the Secrets of Language: An Psychological Journey through Redox Reaction Exercise And Answers

In a digitally-driven world where monitors reign supreme and instant transmission drowns out the subtleties of language, the profound secrets and psychological nuances concealed within phrases often get unheard. However, situated within the pages of **Redox Reaction Exercise And Answers** a fascinating literary prize pulsing with organic emotions, lies an extraordinary journey waiting to be undertaken. Published by a skilled wordsmith, that wonderful opus invites visitors on an introspective journey, lightly unraveling the veiled truths and profound affect resonating within the very cloth of each word. Within the mental depths of the emotional review, we can embark upon a sincere exploration of the book is key themes, dissect its captivating publishing design, and fail to the effective resonance it evokes heavy within the recesses of readers hearts.

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