

Answer Key

Unit 1 Biochemistry

Answers to Unit 1 Preparation Questions Assessing Student Readiness (Student textbook pages 4–7)

- Sketches should depict a central atomic nucleus surrounded by four valence electrons (or six electrons in total), or six neutrons and six protons within the nucleus.
- b
- a
- metal, non-metal ion
- Since the atoms in a glucose molecule share valence electrons, the bonds between these atoms must be covalent.
 - Glucose is a stable molecule because its covalent bonds are stable and strong.
- water—molecular, atoms (both non-metals) are joined by covalent bonds
 - carbon dioxide—molecular, atoms (both non-metals) are joined by covalent bonds
 - glucose—molecular, atoms (two non-metals and uncharged hydrogen) are joined by covalent bonds
 - sodium chloride—ionic, ions (a metal and a non-metal) joined by an ionic bond
 - calcium phosphate—ionic, ions (a metal and a polyatomic ion) joined by ionic bonds
 - methane—molecular, atoms (a non-metal and uncharged hydrogen) joined by covalent bonds
 - oxygen—molecule or molecular element, atoms joined by double covalent bond, but not a molecular compound because it is formed from only one element (two oxygen atoms)
 - ammonia—molecular, atoms (a non-metal and uncharged hydrogen) joined by covalent bonds
- chloride
 - sulphate/sulfate
 - magnesium ion
 - copper (II) ion
- OH^-
 - S^{2-}
 - Al^{3+}
 - Fe^{3+}
- H_2O or water
- $\text{water} \rightarrow \text{hydrogen} + \text{oxygen}$
- d
- sodium hydroxide + hydrochloric acid $\rightarrow$
water + sodium chloride
- Since the pH of the solution increased, the student must have added a base to the solution.
 - One of: When the second substance was added, it released hydroxide ions (OH^-), which combined with protons (H^+) to produce water. The second substance accepted protons (H^+) from the solution.
- decomposition
 - neutralization
 - decomposition
 - synthesis
 - neutralization
 - synthesis
 - decomposition
 - synthesis
 - synthesis
 - synthesis
- pH 7
- c
- One of: enzymes, cell structure, membrane transport, muscle
- A molecule is a substance composed of two or more atoms that are bonded together covalently. A macromolecule is a large, complex molecule. Usually, the subunits of a macromolecule are small molecules bonded together.
- An enzyme is a macromolecule that speeds up chemical reactions.
- c
- c
- b

Review Unit 1 Biochemistry Answers

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molecules regulation of enzymes and reaction sequences lab techniques for purification of proteins and nucleic acids and lab techniques of molecular genetics Also each unit contains more applications to biological systems The text provides a well organized and rigorous approach suitable for classroom use or self instruction Each unit begins with a 1 to 2 page presentation of basic concepts followed by about 20 questions and problems with sample responses Self tests appear after every 2 to 3 units and there is a cumulative self test at the end of the book **Nuclear Science Abstracts** ,1973-07

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