

Percent, Actual, and Theoretical Yield



a) I began this reaction with 20 grams of lithium hydroxide. What is my theoretical yield of lithium chloride?

b) I actually produced 6 grams of lithium chloride. What is my percent yield?



a) If I start with 5 grams of C_3H_8 , what is my theoretical yield of water?

b) I got a percent yield of 75%. How many grams of water did I make?



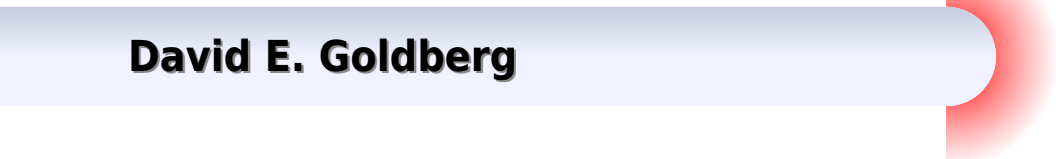
My theoretical yield of beryllium chloride was 10.7 grams. If my actual yield was 4.5 grams, what was my percent yield?



What is my theoretical yield of sodium oxide if I start with 20 grams of calcium oxide?

Percent Actual And Theoretical Yield Answer Key

David E. Goldberg



Percent Actual And Theoretical Yield Answer Key:

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chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students

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